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(54) Title: BIOMEDICAL DEVICES CONTAINING INTERNAL WETTING AGENTS

(57) Abstract: This invention includes a wettable biomedical device containing a high molecular weight hydrophilic polymer and a hydroxyl-functionalized silicone-containing monomer.

WO 03/022321 A2

## BIOMEDICAL DEVICES CONTAINING INTERNAL WETTING AGENTS

## RELATED PATENT APPLICATIONS

This patent application claims priority of a provisional application, U.S.  
5 Ser. No. 60/318,536 which was filed on September 10, 2001.

## FIELD OF THE INVENTION

This invention relates to silicone hydrogels that contain internal wetting  
agents, as well as methods for their production and use.

10

## BACKGROUND OF THE INVENTION

Contact lenses have been used commercially to improve vision since  
at least the 1950s. The first contact lenses were made of hard materials and  
as such were somewhat uncomfortable to users. Modern lenses have been  
15 developed that are made of softer materials, typically hydrogels and  
particularly silicone hydrogels. Silicone hydrogels are water-swollen polymer  
networks that have high oxygen permeability and surfaces that are more  
hydrophobic than hydrophilic. These lenses provide a good level of comfort to  
many lens wearers, but there are some users who experience discomfort and  
20 excessive ocular deposits leading to reduced visual acuity when using these  
lenses. This discomfort and deposits has been attributed to the hydrophobic  
character of the surfaces of lenses and the interaction of those surfaces with  
the protein, lipids and mucin and the hydrophilic surface of the eye.

Others have tried to alleviate this problem by coating the surface of  
25 silicone hydrogel contact lenses with hydrophilic coatings, such as plasma  
coatings. Uncoated lenses having low incidences of surface deposits are not  
disclosed.

Incorporating internal hydrophilic agents (or wetting agents) into a  
macromer containing reaction mixture has been disclosed. However, not all  
30 silicone containing macromers display compatibility with hydrophilic polymers.  
Modifying the surface of a polymeric article by adding polymerizable  
surfactants to a monomer mix used to form the article has also been  
disclosed. However, lasting *in vivo* improvements in wettability and  
reductions in surface deposits are not likely.

Polyvinylpyrrolidone (PVP) or poly-2-ethyl-2-oxazoline have been added to a hydrogel composition to form an interpenetrating network which shows a low degree of surface friction, a low dehydration rate and a high degree of biodeposit resistance. However, the hydrogel formulations disclosed are conventional hydrogels and there is no disclosure on how to incorporate hydrophobic components, such as siloxane monomers, without losing monomer compatibility.

While it may be possible to incorporate high molecular weight polymers as internal wetting agents into silicone hydrogel lenses, such polymers are difficult to solubilize in reaction mixtures which contain silicones. In order to solubilize these wetting agents, silicone macromers or other prepolymers must be used. These silicone macromers or prepolymers must be prepared in a separate step and then subsequently mixed with the remaining ingredients of the silicone hydrogel formulation. This additional step (or steps) increases the cost and the time it takes to produce these lenses. Moreover, these approaches have failed to produce a ophthalmic device which is sufficiently wettable to allow its use as a contact lens without a coating.

Therefore it would be advantageous to find a lens formulation that does not require the use of silicone macromers or other prepolymers and is suitable for extended wear without a surface treatment.

## SUMMARY OF THE INVENTION

The present invention relates to wettable silicone hydrogels formed from a reaction mixture comprising, consisting essentially of or consisting of at least one high molecular weight hydrophilic polymer and at least one hydroxyl-functionalized silicone-containing monomer.

The present invention further relates to biomedical devices formed from a reaction mixture comprising, consisting essentially of, or consisting of a high molecular weight hydrophilic polymer and an effective amount of an hydroxyl-functionalized silicone-containing monomer.

The present invention further relates to a method of preparing a biomedical device comprising, consisting essentially of or consisting of mixing a high molecular weight hydrophilic polymer and an effective amount of a

hydroxyl-functionalized silicone-containing monomer to form a clear solution, and curing said solution.

The present invention yet further relates to a method comprising, consisting essentially of or consisting of the steps of (a) mixing a high  
5 molecular weight hydrophilic polymer and an effective amount of an hydroxyl-functionalized silicone-containing monomer and (b) curing the product of step (a) to form a biomedical device.

The present invention yet further relates to a method comprising, consisting essentially of or consisting of the steps of (a) mixing a high  
10 molecular weight hydrophilic polymer and an effective amount of a hydroxyl-functionalized silicone containing monomer and (b) curing the product of step (a) at or above a minimum gel time, to form a wettable biomedical device.

The present invention still further relates to a method for improving the wettability of an ophthalmic device formed from a reaction mixture comprising,  
15 consisting essentially of and consisting of adding at least one high molecular weight hydrophilic polymer and an effective amount of at least one compatibilizing monomer to said reaction mixture.

The present invention still further relates to a method for improving the wettability of an ophthalmic device formed from a reaction mixture comprising,  
20 consisting essentially of and consisting of adding at least one high molecular weight hydrophilic polymer and an effective amount of at least one hydroxyl-functionalized silicone containing monomer to said reaction mixture.

The present invention still further relates to a biomedical device formed from a reaction mixture comprising, consisting essentially of and consisting of  
25 at least one hydroxyl-functionalized silicone-containing monomer and an amount of high molecular weight hydrophilic polymer sufficient to provide said device, without a surface treatment, with an advancing contact angle of less than about 80°, less than about 70° or less than about 60°.

The present invention still further relates to an ophthalmic device  
30 formed from a reaction mixture comprising, consisting essentially of or consisting of at least one hydroxyl-functionalized silicone-containing monomer and an amount of high molecular weight hydrophilic polymer sufficient to provide said device, without a surface treatment, with a tear film break up time

after about one day of wear of at least about 7 seconds or equal to or greater than tear film break up time for an ACUVUE® contact lens.

A device comprising a silicone hydrogel contact lens which is substantially free from surface deposition without surface modification

5

## DETAILED DESCRIPTION OF THE INVENTION

It has been surprisingly found that biomedical devices, and particularly ophthalmic devices having exceptional in vivo or clinical wettability, without surface modification, may be made by including an effective amount of a high molecular weight hydrophilic polymer and an effective amount of a hydroxyl-functionalized silicone-containing monomer in a silicone hydrogel formulation. By exceptional wettability we mean a decrease in advancing dynamic contact angle of at least about 10% and preferably at least about 20% in some embodiments at least about 50% as compared to a similar formulation without any hydrophilic polymer. Prior to the present invention ophthalmic devices formed from silicone hydrogels either had to be surface modified to provide clinical wettability or be formed from at least one silicone containing macromer having hydroxyl functionality.

As used herein, a "biomedical device" is any article that is designed to be used while either in or on mammalian tissues or fluid, and preferably in or on human tissue or fluids. Examples of these devices include but are not limited to catheters, implants, stents, and ophthalmic devices such as intraocular lenses and contact lenses. The preferred biomedical devices are ophthalmic devices, particularly contact lenses, most particularly contact lenses made from silicone hydrogels.

As used herein, the terms "lens" and "ophthalmic device" refer to devices that reside in or on the eye. These devices can provide optical correction, wound care, drug delivery, diagnostic functionality, cosmetic enhancement or effect or a combination of these properties. The term lens includes but is not limited to soft contact lenses, hard contact lenses, intraocular lenses, overlay lenses, ocular inserts, and optical inserts.

As used herein the term "monomer" is a compound containing at least one polymerizable group and an average molecular weight of about less than 2000 Daltons, as measure via gel permeation chromatography refractive

index detection. Thus, monomers, include dimers and in some cases oligomers, including oligomers made from more than one monomeric unit.

As used herein, the phrase "without a surface treatment" means that the exterior surfaces of the devices of the present invention are not separately  
5 treated to improve the wettability of the device. Treatments which may be foregone because of the present invention include, plasma treatments, grafting, coating and the like. However, coatings which provide properties other than improved wettability, such as, but not limited to antimicrobial coatings may be applied to devices of the present invention.

10 Various molecular weight ranges are disclosed herein. For compounds having discrete molecular structures, the molecular weights reported herein are calculated based upon the molecular formula and reported in gm/mol. For polymers molecular weights (number average) are measured via gel permeation chromatography refractive index detection and reported in Daltons  
15 or are measured via kinematic viscosity measurements, as described in Encyclopedia of Polymer Science and Engineering, N-Vinyl Amide Polymers, Second edition, Vol 17, pgs. 198-257, John Wiley & Sons Inc. and reported in K-values.

#### 20 High Molecular Weight Hydrophilic Polymer

As used herein, "high molecular weight hydrophilic polymer" refers to substances having a weight average molecular weight of no less than about 100,000 Daltons, wherein said substances upon incorporation to silicone hydrogel formulations, increase the wettability of the cured silicone hydrogels.

25 The preferred weight average molecular weight of these high molecular weight hydrophilic polymers is greater than about 150,000; more preferably between about 150,000 to about 2,000,000 Daltons, more preferably still between about 300,000 to about 1,800,000 Daltons, most preferably about 500,000 to about 1,500,000 Daltons.

30 Alternatively, the molecular weight of hydrophilic polymers of the invention can be also expressed by the K-value, based on kinematic viscosity measurements, as described in Encyclopedia of Polymer Science and Engineering, N-Vinyl Amide Polymers, Second edition, Vol 17, pgs. 198-257, John Wiley & Sons Inc. When expressed in this manner, hydrophilic

monomers having K-values of greater than about 46 and preferably between about 46 and about 150. The high molecular weight hydrophilic polymers are present in the formulations of these devices in an amount sufficient to provide contact lenses, which without surface modification remain substantially free from surface depositions during use. Typical use periods include at least about 8 hours, and preferably worn several days in a row, and more preferably for 24 hours or more without removal. Substantially free from surface deposition means that, when viewed with a slit lamp, at least about 70% and preferably at least about 80%, and more preferably about 90% of the lenses worn in the patient population display depositions rated as none or slight, over the wear period.

Suitable amounts of high molecular weight hydrophilic polymer include from about 1 to about 15 weight percent, more preferably about 3 to about 15 percent, most preferably about 5 to about 12 percent, all based upon the total of all reactive components.

Examples of high molecular weight hydrophilic polymers include but are not limited to polyamides, polylactones, polyimides, poly lactams and functionalized polyamides, polylactones, polyimides, poly lactams, such as DMA functionalized by copolymerizing DMA with a lesser molar amount of a hydroxyl-functional monomer such as HEMA, and then reacting the hydroxyl groups of the resulting copolymer with materials containing radical polymerizable groups, such as isocyanatoethylmethacrylate or methacryloyl chloride. Hydrophilic prepolymers made from DMA or n-vinyl pyrrolidone with glycidyl methacrylate may also be used. The glycidyl methacrylate ring can be opened to give a diol which may be used in conjunction with other hydrophilic prepolymer in a mixed system to increase the compatibility of the high molecular weight hydrophilic polymer, hydroxyl-functionalized silicone containing monomer and any other groups which impart compatibility. The preferred high molecular weight hydrophilic polymers are those that contain a cyclic moiety in their backbone, more preferably, a cyclic amide or cyclic imide. High molecular weight hydrophilic polymers include but are not limited to poly-N-vinyl pyrrolidone, poly-N-vinyl-2- piperidone, poly-N-vinyl-2-caprolactam, poly-N-vinyl-3-methyl-2- caprolactam, poly-N-vinyl-3-methyl-2-piperidone, poly-N-vinyl-4-methyl-2- piperidone, poly-N-vinyl-4-methyl-2-

caprolactam, poly-N-vinyl-3-ethyl-2- pyrrolidone, and poly-N-vinyl-4,5-dimethyl-2-pyrrolidone, polyvinylimidazole, poly-N-N-dimethylacrylamide, polyvinyl alcohol, polyacrylic acid, polyethylene oxide, poly 2 ethyl oxazoline, heparin polysaccharides, polysaccharides, mixtures and copolymers  
5 (including block or random, branched, multichain, comb-shaped or star shaped) thereof where poly-N-vinylpyrrolidone (PVP) is particularly preferred. Copolymers might also be used such as graft copolymers of PVP.

The high molecular weight hydrophilic polymers provide improved wettability, and particularly improved in vivo wettability to the medical devices  
10 of the present invention. Without being bound by any theory, it is believed that the high molecular weight hydrophilic polymers are hydrogen bond receivers which in aqueous environments, hydrogen bond to water, thus becoming effectively more hydrophilic. The absence of water facilitates the incorporation of the hydrophilic polymer in the reaction mixture. Aside from  
15 the specifically named high molecular weight hydrophilic polymers, it is expected that any high molecular weight polymer will be useful in this invention provided that when said polymer is added to a silicone hydrogel formulation, the hydrophilic polymer (a) does not substantially phase separate from the reaction mixture and (b) imparts wettability to the resulting cured  
20 polymer. In some embodiments it is preferred that the high molecular weight hydrophilic polymer be soluble in the diluent at processing temperatures. Manufacturing processes which use water or water soluble diluents may be preferred due to their simplicity and reduced cost. In these embodiments high molecular weight hydrophilic polymers which are water soluble at processing  
25 temperatures are preferred.

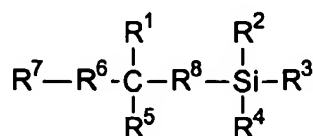
#### Hydroxyl-functionalized silicone containing monomer

As used herein a "hydroxyl-functionalized silicone containing monomer" is a compound containing at least one polymerizable group having an average  
30 molecular weight of about less than 5000 Daltons as measured via gel permeation chromatography, refractive index detection, and preferably less than about 3000 Daltons, which is capable of compatibilizing the silicone containing monomers included in the hydrogel formulation with the hydrophilic polymer. Hydroxyl functionality is very efficient at improving hydrophilic

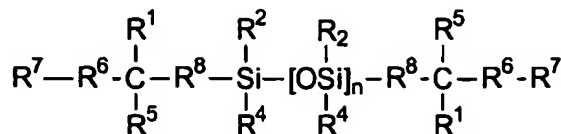
compatibility. Thus, in a preferred embodiment hydroxyl-functionalized silicone containing monomers of the present invention comprise at least one hydroxyl group and at least one "-Si-O-Si-"group. It is preferred that silicone and its attached oxygen account for more than about 10 weight percent of said hydroxyl-functionalized silicone containing monomer, more preferably more than about 20 weight percent.

The ratio of Si to OH in the hydroxyl-functionalized silicone containing monomer is also important to providing a hydroxyl functionalized silicone containing monomer which will provide the desired degree of compatibilization. If the ratio of hydrophobic portion to OH is too high, the hydroxyl-functionalized silicone monomer may be poor at compatibilizing the hydrophilic polymer, resulting in incompatible reaction mixtures. Accordingly, in some embodiments, the Si to OH ratio is less than about 15:1, and preferably between about 1:1 to about 10:1. In some embodiments primary alcohols have provided improved compatibility compared to secondary alcohols. Those of skill in the art will appreciate that the amount and selection of hydroxyl-functionalized silicone containing monomer will depend on how much hydrophilic polymer is needed to achieve the desired wettability and the degree to which the silicone containing monomer is incompatible with the hydrophilic polymer.

Examples of hydroxyl-functionalized silicone containing monomers include monomers of Formulae I and II



I



II

wherein:

n is an integer between 3 and 35, and preferably between 4 and 25;

R<sup>1</sup> is hydrogen, C<sub>1-6</sub>alkyl;

R<sup>2</sup>, R<sup>3</sup>, and R<sup>4</sup>, are independently, C<sub>1-6</sub>alkyl, triC<sub>1-6</sub>alkylsiloxyl, phenyl,

5 naphthyl, substituted C<sub>1-6</sub>alkyl, substituted phenyl, or substituted naphthyl

where the alkyl substituents are selected from one or more members of the group consisting of C<sub>1-6</sub>alkoxycarbonyl, C<sub>1-6</sub>alkyl, C<sub>1-6</sub>alkoxy,

amide, halogen, hydroxyl, carboxyl, C<sub>1-6</sub>alkylcarbonyl and formyl, and

where the aromatic substituents are selected from one or more

10 members of the group consisting of C<sub>1-6</sub>alkoxycarbonyl, C<sub>1-6</sub>alkyl,

C<sub>1-6</sub>alkoxy, amide, halogen, hydroxyl, carboxyl, C<sub>1-6</sub>alkylcarbonyl and formyl;

R<sup>5</sup> is hydroxyl, an alkyl group containing one or more hydroxyl groups; or

(CH<sub>2</sub>(CR<sup>9</sup>R<sup>10</sup>)<sub>y</sub>O)<sub>x</sub>-R<sup>11</sup> wherein y is 1 to 5, preferably 1 to 3, x is an integer of

15 1 to 100, preferably 2 to 90 and more preferably 10 to 25; R<sup>9</sup> - R<sup>11</sup> are

independently selected from H, alkyl having up to 10 carbon atoms and alkyls having up to 10 carbon atoms substituted with at least one polar functional group,

20 R<sup>6</sup> is a divalent group comprising up to 20 carbon atoms;

R<sup>7</sup> is a monovalent group that can under free radical and/or cationic polymerization and comprising up to 20 carbon atoms

R<sup>8</sup> is a divalent or trivalent group comprising up to 20 carbon atoms.

25 Reaction mixtures of the present invention may include more than one hydroxyl-functionalized silicone containing monomer.

For monofunctional hydroxyl functionalized silicone containing

monomer the preferred R<sup>1</sup> is hydrogen, and the preferred R<sup>2</sup>, R<sup>3</sup>, and R<sup>4</sup>, are C<sub>1-6</sub>alkyl and triC<sub>1-6</sub>alkylsiloxyl, most preferred methyl and trimethylsiloxyl. For

30 multifunctional (difunctional or higher) R<sup>1</sup>-R<sup>4</sup> independently comprise

ethylenically unsaturated polymerizable groups and more preferably comprise an acrylate, a styryl, a C<sub>1-6</sub>alkylacrylate, acrylamide, C<sub>1-6</sub>alkylacrylamide, N-vinyl lactam, N-vinylamide, , C<sub>2-12</sub>alkenyl, C<sub>2-12</sub>alkenylphenyl, C<sub>2-12</sub>alkenyl naphthyl, or C<sub>2-6</sub>alkenylphenylC<sub>1-6</sub>alkyl.

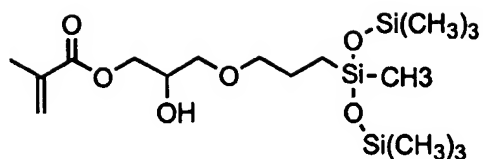
The preferred  $R^5$  is hydroxyl,  $-CH_2OH$  or  $CH_2CHOHCH_2OH$ , with hydroxyl being most preferred.

- The preferred  $R^6$  is a divalent  $C_{1-6}$ alkyl,  $C_{1-6}$ alkyloxy,  $C_{1-6}$ alkyloxy $C_{1-6}$ alkyl, phenylene, naphthalene,  $C_{1-12}$ cycloalkyl,
- 5  $C_{1-6}$ alkoxycarbonyl, amide, carboxy,  $C_{1-6}$ alkylcarbonyl, carbonyl,  $C_{1-6}$ alkoxy, substituted  $C_{1-6}$ alkyl, substituted  $C_{1-6}$ alkyloxy, substituted  $C_{1-6}$ alkyloxy $C_{1-6}$ alkyl, substituted phenylene, substituted naphthalene, substituted  $C_{1-12}$ cycloalkyl, where the substituents are selected from one or more members of the group consisting of  $C_{1-6}$ alkoxycarbonyl,  $C_{1-6}$ alkyl,  $C_{1-6}$ alkoxy, amide, halogen,
- 10 hydroxyl, carboxyl,  $C_{1-6}$ alkylcarbonyl and formyl. The particularly preferred  $R^6$  is a divalent methyl (methylene).

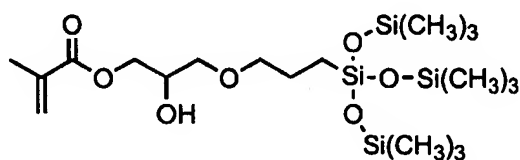
- The preferred  $R^7$  comprises a free radical reactive group, such as an acrylate, a styryl, vinyl, vinyl ether, itaconate group, a  $C_{1-6}$ alkylacrylate, acrylamide,  $C_{1-6}$ alkylacrylamide, N-vinylactam, N-vinylamide, ,  $C_{2-12}$ alkenyl,
- 15  $C_{2-12}$ alkenylphenyl,  $C_{2-12}$ alkenylnaphthyl, or  $C_{2-6}$ alkenylphenyl $C_{1-6}$ alkyl or a cationic reactive group such as vinyl ether or epoxide groups. The particularly preferred  $R^7$  is methacrylate.

- The preferred  $R^8$  is a divalent  $C_{1-6}$ alkyl,  $C_{1-6}$ alkyloxy,  $C_{1-6}$ alkyloxy $C_{1-6}$ alkyl, phenylene, naphthalene,  $C_{1-12}$ cycloalkyl,
- 20  $C_{1-6}$ alkoxycarbonyl, amide, carboxy,  $C_{1-6}$ alkylcarbonyl, carbonyl,  $C_{1-6}$ alkoxy, substituted  $C_{1-6}$ alkyl, substituted  $C_{1-6}$ alkyloxy, substituted  $C_{1-6}$ alkyloxy $C_{1-6}$ alkyl, substituted phenylene, substituted naphthalene, substituted  $C_{1-12}$ cycloalkyl, where the substituents are selected from one or more members of the group consisting of  $C_{1-6}$ alkoxycarbonyl,  $C_{1-6}$ alkyl,  $C_{1-6}$ alkoxy, amide, halogen,
- 25 hydroxyl, carboxyl,  $C_{1-6}$ alkylcarbonyl and formyl. The particularly preferred  $R^8$  is  $C_{1-6}$ alkyloxy $C_{1-6}$ alkyl.

- Examples of hydroxyl-functionalized silicone containing monomer of Formula I that are particularly preferred are 2-propenoic acid, 2-methyl-, 2-hydroxy-3-[3-[1,3,3,3-tetramethyl-1-
- 30 [(trimethylsilyl)oxy]disiloxanyl]propoxy]propyl ester (which can also be named (3-methacryloxy-2-hydroxypropyloxy)propylbis(trimethylsiloxy)methylsilane)

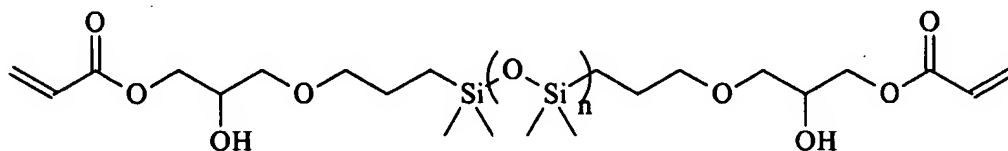


- The above compound, (3-methacryloxy-2-hydroxypropyloxy)propylbis(trimethylsiloxy)methylsilane is formed from an epoxide, which produces an 80:20 mixture of the compound shown above and (2-methacryloxy-3-hydroxypropyloxy)propylbis(trimethylsiloxy)methylsilane. In some embodiments of the present invention it is preferred to have some amount of the primary hydroxyl present, preferably greater than about 10 wt% and more preferably at least about 20 wt%.
- Other suitable hydroxyl-functionalized silicone containing monomers include (3-methacryloxy-2-hydroxypropyloxy)propyltris(trimethylsiloxy)silane



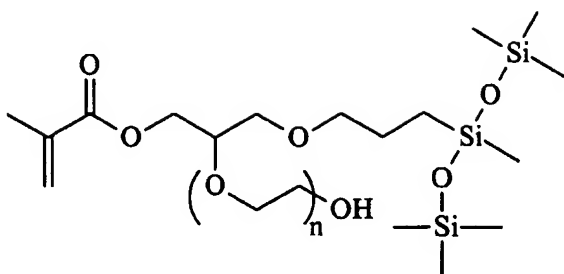
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bis-3-methacryloxy-2-hydroxypropyloxypropyl polydimethylsiloxane



20

3-methacryloxy-2-(2-hydroxyethoxy)propyloxy)propylbis(trimethylsiloxy)methylsilane

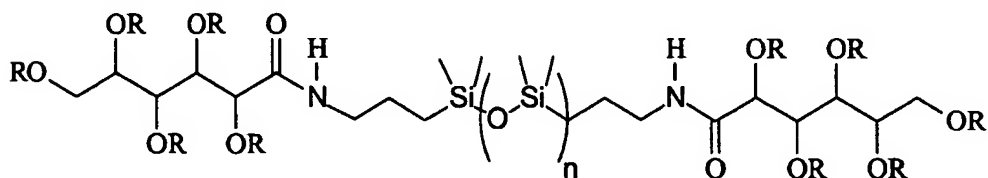


- 5 N,N,N',N'-tetrakis(3-methacryloxy-2-hydroxypropyl)- $\alpha,\omega$ -bis-3-aminopropyl-polydimethylsiloxane

10

The reaction products of glycidyl methacrylate with amino-functional polydimethylsiloxanes may also be used as a hydroxyl-functional silicone containing monomer. Other suitable hydroxyl-functional silicone containing monomers include those disclosed in columns 6,7 and 8 of US 5,994,488, and  
 15 monomers disclosed in 4,259,467; 4,260,725; 4,261,875; 4,649,184; 4,139,513, 4,139,692, US 2002/0016383, 4,139,513 and 4,139,692. These and any other patents or applications cited herein are incorporated by reference.

20 Still additional structures which may be suitable hydroxyl-functionalized silicone containing monomers include those similar to the compounds disclosed in Pro. ACS Div. Polym. Mat. Sci. Eng., April 13-17, 1997, p. 42, and having the following structure:



where  $n = 1-50$  and R independently comprise H or a polymerizable unsaturated group, with at least one R comprising a polymerizable group, and at least one R, and preferably 3-8 R, comprising H.

Additional suitable hydroxyl-functionalized silicone containing  
5 monomers are disclosed in U.S. Patent 4,235,985.

These components may be removed from the hydroxyl-functionalized monomer via known methods such as liquid phase chromatography, distillation, recrystallization or extraction, or their formation may be avoided by careful selection of reaction conditions and reactant ratios.

10 Suitable monofunctional hydroxyl-functionalized silicone monomers are commercially available from Gelest, Inc. Morrisville, PA. Suitable multifunctional hydroxyl-functionalized silicone monomers are commercially available from Gelest, Inc, Morrisville, PA or may be made using the procedures disclosed in 5,994,488 and 5,962,548. Suitable PEG type  
15 hydroxyl-functionalized silicone monomers may be made using the procedures disclosed in PCT/JP02/02231.

While hydroxyl-functionalized silicone containing monomers have been found to be particularly suitable for providing compatible polymers for biomedical devices, and particularly ophthalmic devices, any functionalized  
20 silicone containing monomer which, when polymerized and/or formed into a final article is compatible with the selected hydrophilic components may be used. Suitable functionalized silicone containing monomers may be selected using the following monomer compatibility test. In this test one gram of each of mono-3-methacryloxypropyl terminated, mono-butyl terminated  
25 polydimethylsiloxane (mPDMS MW 800-1000) and a monomer to be tested are mixed together in one gram of 3,7-dimethyl-3-octanol at about 20°C. A mixture of 12 weight parts K-90 PVP and 60 weight parts DMA is added drop-wise to hydrophobic component solution, with stirring, until the solution remains cloudy after three minutes of stirring. The mass of the added blend  
30 of PVP and DMA is determined in grams and recorded as the monomer compatibility index. Any hydroxyl-functionalized silicone-containing monomer having a compatibility index of greater than 0.2 grams, more preferably

greater than about 0.7 grams and most preferably greater than about 1.5 grams will be suitable for use in this invention.

An "effective amount" or a "compatibilizing effective amount" of the hydroxyl-functionalized silicone-containing monomers of the invention is the amount needed to compatibilize or dissolve the high molecular weight hydrophilic polymer and the other components of the polymer formulation. Thus, the amount of hydroxyl-functional silicone containing monomer will depend in part on the amount of hydrophilic polymer which is used, with more hydroxyl-functionalized silicone containing monomer being needed to compatibilize higher concentrations of hydrophilic polymer. Effective amounts of hydroxyl-functionalized silicone containing monomer in the polymer formulation include about 5% (weight percent, based on the weight percentage of the reactive components) to about 90 %, preferably about 10% to about 80%, most preferably, about 20% to about 50%.

In addition to the high molecular weight hydrophilic polymers and the hydroxyl-functionalized silicone containing monomers of the invention other hydrophilic and hydrophobic monomers, crosslinkers, additives, diluents, polymerization initiators may be used to prepare the biomedical devices of the invention. In addition to high molecular weight hydrophilic polymer and hydroxyl-functionalized silicone containing monomer, the hydrogel formulations may include additional silicone containing monomers, hydrophilic monomers, and cross linkers to give the biomedical devices of the invention.

#### Additional Silicone Containing Monomers

With respect to the additional silicone containing monomers, amide analogs of TRIS described in U.S. 4,711,943, vinylcarbamate or carbonate analogs described in U.S. Pat. 5,070,215, and siloxane containing monomers contained in U.S. Pat. 6,020,445 are useful and these aforementioned patents as well as any other patents mentioned in this specification are hereby incorporated by reference. More specifically, 3-methacryloxypropyltris(trimethylsiloxy)silane (TRIS), monomethacryloxypropyl terminated polydimethylsiloxanes, polydimethylsiloxanes, 3-methacryloxypropylbis(trimethylsiloxy)methylsilane, methacryloxypropylpentamethyl disiloxane and combinations thereof are

particularly useful as additional silicone-containing monomers of the invention. Additional silicone containing monomers may be present in amounts of about 0 to about 75 wt%, more preferably of about 5 and about 60 and most preferably of about 10 and 40 weight%.

5

#### Hydrophilic Monomers

Additionally, reaction components of the present invention may also include any hydrophilic monomers used to prepare conventional hydrogels. For example monomers containing acrylic groups ( $\text{CH}_2=\text{CRCOX}$ , where R is hydrogen or  $\text{C}_{1-6}$ alkyl and X is O or N) or vinyl groups ( $-\text{C}=\text{CH}_2$ ) may be used. Examples of additional hydrophilic monomers are N,N-dimethylacrylamide, 2-hydroxyethyl methacrylate, glycerol monomethacrylate, 2-hydroxyethyl methacrylamide, polyethyleneglycol monomethacrylate, methacrylic acid, acrylic acid, N-vinyl pyrrolidone, N-vinyl-N-methyl acetamide, N-vinyl-N-ethyl acetamide, N-vinyl-N-ethyl formamide, N-vinyl formamide and combinations thereof.

Aside the additional hydrophilic monomers mentioned above, polyoxyethylene polyols having one or more of the terminal hydroxyl groups replaced with a functional group containing a polymerizable double bond may be used. Examples include polyethylene glycol, as disclosed in US 5,484,863, ethoxylated alkyl glucoside, as disclosed in US 5,690,953, US 5,304,584, and ethoxylated bisphenol A, as disclosed in US 5,565,539, reacted with one or more molar equivalents of an end-capping group such as isocyanatoethyl methacrylate, methacrylic anhydride, methacryloyl chloride, vinylbenzoyl chloride, and the like, produce a polyethylene polyol having one or more terminal polymerizable olefinic groups bonded to the polyethylene polyol through linking moieties such as carbamate, urea or ester groups.

Still further examples include the hydrophilic vinyl carbonate or vinyl carbamate monomers disclosed in U.S. Pat. Nos. 5,070,215, the hydrophilic oxazolone monomers disclosed in U.S. Pat. No. 4,910,277, the disclosures of which are incorporated herein by reference and polydextran.

The preferred additional hydrophilic monomers are N,N-dimethylacrylamide (DMA), 2-hydroxyethyl methacrylate (HEMA), glycerol methacrylate, 2-hydroxyethyl methacrylamide, N-vinylpyrrolidone (NVP),

polyethyleneglycol monomethacrylate, methacrylic acid, acrylic acid and combinations thereof, with hydrophilic monomers comprising DMA being particularly preferred. Additional hydrophilic monomers may be present in amounts of about 0 to about 70 wt%, more preferably of about 5 and about 60 and most preferably of about 10 and 50 weight%.

### Crosslinkers

Suitable crosslinkers are compounds with two or more polymerizable functional groups. The crosslinker may be hydrophilic or hydrophobic and in some embodiments of the present invention mixtures of hydrophilic and hydrophobic crosslinkers have been found to provide silicone hydrogels with improved optical clarity (reduced haziness compared to a CSI Thin Lens). Examples of suitable hydrophilic crosslinkers include compounds having two or more polymerizable functional groups, as well as hydrophilic functional groups such as polyether, amide or hydroxyl groups. Specific examples include TEGDMA (tetraethyleneglycol dimethacrylate), TrEGDMA (triethyleneglycol dimethacrylate), ethyleneglycol dimethacrylate (EGDMA), ethylenediamine dimethacrylamide, glycerol dimethacrylate and combinations thereof. Examples of suitable hydrophobic crosslinkers include multifunctional hydroxyl-functionalized silicone containing monomer, multifunctional polyether-polydimethylsiloxane block copolymers, combinations thereof and the like. Specific hydrophobic crosslinkers include acryloxypropyl terminated polydimethylsiloxane ( $n = 10$  or  $20$ ) (acPDMS), hydroxylacrylate functionalized siloxane macromer, methacryloxypropyl terminated PDMS, butanediol dimethacrylate, divinyl benzene, 1,3-bis(3-methacryloxypropyl)tetrakis(trimethylsiloxy)disiloxane and mixtures thereof. Preferred crosslinkers include TEGDMA, EGDMA, acPDMS and combinations thereof. The amount of hydrophilic crosslinker used is generally about 0 to about 2 weight% and preferably from about 0.5 to about 2 weight % and the amount of hydrophobic crosslinker is about 0 to about 5 weight %, which can alternatively be referred to in mol% of about 0.01 to about 0.2 mmole/gm reactive components, preferably about 0.02 to about 0.1 and more preferably 0.03 to about 0.6 mmole/gm.

Increasing the level of crosslinker in the final polymer has been found to reduce the amount of haze. However, as crosslinker concentration increases above about 0.15 mmole/gm reactive components modulus increases above generally desired levels (greater than about 90 psi). Thus, in  
5 the present invention the crosslinker composition and amount is selected to provide a crosslinker concentration in the reaction mixture of between about 0.01 and about 0.1 mmoles/gm crosslinker.

Additional components or additives, which are generally known in the art may also be included. Additives include but are not limited to ultra-violet  
10 absorbing compounds and monomer, reactive tints, antimicrobial compounds, pigments, photochromic, release agents, combinations thereof and the like.

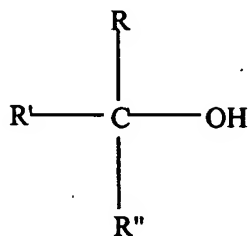
Additional components include other oxygen permeable components such as carbon-carbon triple bond containing monomers and fluorine containing monomers which are known in the art and include fluorine-  
15 containing (meth)acrylates, and more specifically include, for example, fluorine-containing C<sub>2</sub>-C<sub>12</sub> alkyl esters of (meth)acrylic acid such as 2,2,2-trifluoroethyl (meth)acrylate, 2,2,2,2',2',2'-hexafluoroisopropyl (meth)acrylate, 2,2,3, 3,4,4,4-heptafluorobutyl (meth)acrylate, 2,2,3,3,4,4,5,5,6,6,7,7,8,8,8-pentadecafluorooctyl (meth)acrylate, 2,2,3,3,4,4,5,5,6,6,7,7,8,8,9,9-  
20 hexadecafluorononyl (meth)acrylate and the like

### Diluents

The reaction components (hydroxyl-functionalized silicone containing monomer, hydrophilic polymer, crosslinker(s) and other components) are  
25 generally mixed and reacted in the absence of water and optionally, in the presence of at least one diluent to form a reaction mixture. The type and amount of diluent used also effects the properties of the resultant polymer and article. The haze and wettability of the final article may be improved by selecting relatively hydrophobic diluents and/or decreasing the concentration  
30 of diluent used. As discussed above, increasing the hydrophobicity of the diluent may also allow poorly compatible components (as measured by the compatibility test) to be processed to form a compatible polymer and article. However, as the diluent becomes more hydrophobic, processing steps necessary to replace the diluent with water will require the use of solvents

other than water. This may undesirably increase the complexity and cost of the manufacturing process. Thus, it is important to select a diluent which provides the desired compatibility to the components with the necessary level of processing convenience. Diluents useful in preparing the devices of this invention include ethers, esters, alkanes, alkyl halides, silanes, amides, alcohols and combinations thereof. Amides and alcohols are preferred diluents, and secondary and tertiary alcohols are most preferred alcohol diluents. Examples of ethers useful as diluents for this invention include tetrahydrofuran, tripropylene glycol methyl ether, dipropylene glycol methyl ether, ethylene glycol n-butyl ether, diethylene glycol n-butyl ether, diethylene glycol methyl ether, ethylene glycol phenyl ether, propylene glycol methyl ether, propylene glycol methyl ether acetate, dipropylene glycol methyl ether acetate, propylene glycol n-propyl ether, dipropylene glycol n-propyl ether, tripropylene glycol n-butyl ether, propylene glycol n-butyl ether, dipropylene glycol n-butyl ether, tripropylene glycol n-butyl ether, propylene glycol phenyl ether dipropylene glycol dimethyl ether, polyethylene glycols, polypropylene glycols and mixtures thereof. Examples of esters useful for this invention include ethyl acetate, butyl acetate, amyl acetate, methyl lactate, ethyl lactate, i-propyl lactate. Examples of alkyl halides useful as diluents for this invention include methylene chloride. Examples of silanes useful as diluents for this invention include octamethylcyclotetrasiloxane.

Examples of alcohols useful as diluents for this invention include those having the formula



wherein R, R' and R'' are independently selected from H, a linear, branched or cyclic monovalent alkyl having 1 to 10 carbons which may optionally be substituted with one or more groups including halogens, ethers, esters, aryls,

amines, amides, alkenes, alkynes, carboxylic acids, alcohols, aldehydes, ketones or the like, or any two or all three of R, R' and R" can together bond to form one or more cyclic structures, such as alkyl having 1 to 10 carbons which may also be substituted as just described, with the proviso that no more than one of R, R' or R" is H.

It is preferred that R, R' and R" are independently selected from H or unsubstituted linear, branched or cyclic alkyl groups having 1 to 7 carbons. It is more preferred that R, R', and R" are independently selected from unsubstituted linear, branched or cyclic alkyl groups having 1 to 7 carbons. In certain embodiments, the preferred diluent has 4 or more, more preferably 5 or more total carbons, because the higher molecular weight diluents have lower volatility, and lower flammability. When one of the R, R' and R" is H, the structure forms a secondary alcohol. When none of the R, R' and R" are H, the structure forms a tertiary alcohol. Tertiary alcohols are more preferred than secondary alcohols. The diluents are preferably inert and easily displaceable by water when the total number of carbons is five or less.

Examples of useful secondary alcohols include 2-butanol, 2-propanol, menthol, cyclohexanol, cyclopentanol and exonorborneol, 2-pentanol, 3-pentanol, 2-hexanol, 3-hexanol, 3-methyl-2-butanol, 2-heptanol, 2-octanol, 2-nonanol, 2-decanol, 3-octanol, norborneol, and the like.

Examples of useful tertiary alcohols include tert-butanol, tert-amyl, alcohol, 2-methyl-2-pentanol, 2,3-dimethyl-2-butanol, 3-methyl-3-pentanol, 1-methylcyclohexanol, 2-methyl-2-hexanol, 3,7-dimethyl-3-octanol, 1-chloro-2-methyl-2-propanol, 2-methyl-2-heptanol, 2-methyl-2-octanol, 2,2-methyl-2-nonanol, 2-methyl-2-decanol, 3-methyl-3-hexanol, 3-methyl-3-heptanol, 4-methyl-4-heptanol, 3-methyl-3-octanol, 4-methyl-4-octanol, 3-methyl-3-nonanol, 4-methyl-4-nonanol, 3-methyl-3-octanol, 3-ethyl-3-hexanol, 3-methyl-3-heptanol, 4-ethyl-4-heptanol, 4-propyl-4-heptanol, 4-isopropyl-4-heptanol, 2,4-dimethyl-2-pentanol, 1-methylcyclopentanol, 1-ethylcyclopentanol, 1-ethylcyclopentanol, 3-hydroxy-3-methyl-1-butene, 4-hydroxy-4-methyl-1-cyclopentanol, 2-phenyl-2-propanol, 2-methoxy-2-methyl-2-propanol, 2,3,4-trimethyl-3-pentanol, 3,7-dimethyl-3-octanol, 2-phenyl-2-butanol, 2-methyl-1-phenyl-2-propanol and 3-ethyl-3-pentanol, and the like.

A single alcohol or mixtures of two or more of the above-listed alcohols or two or more alcohols according to the structure above can be used as the diluent to make the polymer of this invention.

In certain embodiments, the preferred alcohol diluents are secondary and tertiary alcohols having at least 4 carbons. The more preferred alcohol diluents include tert-butanol, tert-amyl alcohol, 2-butanol, 2-methyl-2-pentanol, 2,3-dimethyl-2-butanol, 3-methyl-3-pentanol, 3-ethyl-3-pentanol, 3,7-dimethyl-3-octanol.

Presently, the most preferred diluents are hexanol, heptanol, octanol, nonanol, decanol, tert-butyl alcohol, 3-methyl-3-pentanol, isopropanol, t amyl alcohol, ethyl lactate, methyl lactate, i-propyl lactate, 3,7-dimethyl-3-octanol, dimethyl formamide, dimethyl acetamide, dimethyl propionamide, N methyl pyrrolidinone and mixtures thereof. Additional diluents useful for this invention are disclosed in US patent 6,020,445, which is incorporated herein by reference.

In one embodiment of the present invention the diluent is water soluble at processing conditions and readily washed out of the lens with water in a short period of time. Suitable water soluble diluents include 1-ethoxy-2-propanol, 1-methyl-2-propanol, t-amyl alcohol, tripropylene glycol methyl ether, isopropanol, 1-methyl-2-pyrrolidone, N,N-dimethylpropionamide, ethyl lactate, dipropylene glycol methyl ether, mixtures thereof and the like. The use of a water soluble diluent allows the post molding process to be conducted using water only or aqueous solutions which comprise water as a substantial component.

In one embodiment, the amount of diluent is generally less than about 50 weight % of the reaction mixture and preferably less than about 40% and more preferably between about 10 and about 30%.

The diluent may also comprise additional components such as release agents. Suitable release agents are water soluble and aid in lens deblocking.

The polymerization initiators includes compounds such as lauryl peroxide, benzoyl peroxide, isopropyl percarbonate, azobisisobutyronitrile, and the like, that generate free radicals at moderately elevated temperatures, and photoinitiator systems such as aromatic alpha-hydroxy ketones, alkoxyoxybenzoin, acetophenones, acyl phosphine oxides, and a tertiary

- amine plus a diketone, mixtures thereof and the like. Illustrative examples of photoinitiators are 1-hydroxycyclohexyl phenyl ketone, 2-hydroxy-2-methyl-1-phenyl-propan-1-one, bis(2,6-dimethoxybenzoyl)-2,4,4-trimethylpentyl phosphine oxide (DMBAPO), bis(2,4,6-trimethylbenzoyl)-
- 5 phenylphosphineoxide (Irgacure 819), 2,4,6-trimethylbenzoyldiphenyl phosphine oxide and 2,4,6-trimethylbenzoyl diphenylphosphine oxide, benzoin methyl ester and a combination of camphorquinone and ethyl 4-(N,N-dimethylamino)benzoate. Commercially available visible light initiator systems include Irgacure 819, Irgacure 1700, Irgacure 1800, Irgacure 819,
- 10 Irgacure 1850 (all from Ciba Specialty Chemicals) and Lucirin TPO initiator (available from BASF). Commercially available UV photoinitiators include Darocur 1173 and Darocur 2959 (Ciba Specialty Chemicals). The initiator is used in the reaction mixture in effective amounts to initiate photopolymerization of the reaction mixture, e.g., from about 0.1 to about 2
- 15 parts by weight per 100 parts of reactive monomer. Polymerization of the reaction mixture can be initiated using the appropriate choice of heat or visible or ultraviolet light or other means depending on the polymerization initiator used. Alternatively, initiation can be conducted without a photoinitiator using, for example, e-beam. However, when a photoinitiator is used, the preferred
- 20 initiator is a combination of 1-hydroxycyclohexyl phenyl ketone and bis(2,6-dimethoxybenzoyl)-2,4,4-trimethylpentyl phosphine oxide (DMBAPO), and the preferred method of polymerization initiation is visible light. The most preferred is bis(2,4,6-trimethylbenzoyl)-phenylphosphineoxide (Irgacure 819®).
- 25 The invention further comprises, consists and consists essentially of a silicone hydrogel, biomedical device, ophthalmic device and contact lenses of the formulae shown below:

Wt% components			
HFSCM	HMWHP	SCM	HM
5-90	1-15, 3-15 or 5-12	0	0
10-80	1-15, 3-15 or 5-12	0	0
20-50	1-15, 3-15 or 5-12	0	0
5-90	1-15, 3-15 or 5-12	0-80, 5-60 or 10-40	0-70, 5-60 or 10-50
10-80	1-15, 3-15 or 5-12	0-80, 5-60 or 10-	0-70, 5-60 or 10-

		40	50
20-50	1-15, 3-15 or 5-12	0-80, 5-60 or 10-40	0-70, 5-60 or 10-50

HFSCM is hydroxyl-functionalized silicone containing monomer

HMWHP is high molecular weight hydrophilic polymer

SCM is silicone containing monomer

HM is hydrophilic monomer

5        The weight percents above are based upon all reactive components.

Thus, the present invention includes silicone hydrogel, biomedical device, ophthalmic device and contact lenses having each of the composition listed in the table, which describes ninety possible compositional ranges. Each of the ranges listed above is prefaced by the word "about". The foregoing range combinations are presented with the proviso that the listed components, and any additional components add up to 100 weight%..

A preferred range of the combined silicone-containing monomers (hydroxyl-functionalized silicone-containing and additional silicone-containing monomers) is from about 5 to 99 weight percent, more preferably about 15 to 90 weight percent, and most preferably about 25 to about 80 weight percent of the reaction components. A preferred range of hydroxyl-functionalized silicone-containing monomer is about 5 to about 90 weight percent, preferably about 10 to about 80, and most preferably about 20 to about 50 weight percent. A preferred range of hydrophilic monomer is from about 0 to about 70 weight percent, more preferably about 5 to about 60 weight percent, and most preferably about 10 to about 50 weight percent of the reactive components. A preferred range of high molecular weight hydrophilic polymer is about 1 to about 15 weight percent, more preferably about 3 to about 15 weight percent, and most preferably about 5 to about 12 weight percent. . All of the about weight percents are based upon the total of all reactive components. A preferred range of diluent is from about 0 to about 70 weight percent, more preferably about 0 to about 50 weight percent, and still more preferably about 0 to about 40 weight percent and in some embodiments, most preferably between about 10 and about 30 weight percent, based upon the weight all component in the reactive mixture. The amount of diluent required varies depending on the nature and relative amounts of the reactive components.

In a preferred embodiment, the reactive components comprise 2-propenoic acid, 2-methyl-,2-hydroxy-3-[3-[1,3,3,3-tetramethyl-1-[(trimethylsilyl)oxy]disiloxanyl]propoxy]propyl ester "SiGMA" (~28 wgt. % of the reaction components); (800-1000 MW monomethacryloxypropyl terminated mono-n-butyl terminated polydimethylsiloxane, "mPDMS" (~31%wt); N,N-dimethylacrylamide, "DMA" (~24%wt ); 2-hydroxyethyl methacrylate, "HEMA" (~6%wt); tetraethyleneglycoldimethacrylate, "EGDMA" (~1.5%wt), polyvinylpyrrolidone, "K-90 PVP" (~7%wt); with the balance comprising minor amounts of additives and photoinitiators. The polymerization is most preferably conducted in the presence of about 23% (weight % of the combined monomers and diluent blend) 3,7-dimethyl-3-octanol diluent.

In other preferred embodiments the reactive components comprise those shown in the Table below. All amounts are prefaced by the word "about".

Component	Wt%	
SiGMA	30	30
mPDMS	23	18
DMA	31	31
HEMA	7.5	9
EGDMA	0.75	0.8
PVP	6	6

The polymerizations for the above formulations are preferably conducted in the presence of tert-amyl-alcohol as a diluent comprising about 29 weight percent of the uncured reaction mixture.

#### Processing

The biomedical devices of the invention are prepared by mixing the high molecular weight hydrophilic polymer, the hydroxyl-functionalized silicone-containing monomer, plus one or more of the following: the additional silicone containing monomers, the hydrophilic monomers, the additives ("reactive components"), and the diluents ("reaction mixture"), with a

polymerization initiator and curing by appropriate conditions to form a product that can be subsequently formed into the appropriate shape by lathing, cutting and the like. Alternatively, the reaction mixture may be placed in a mold and subsequently cured into the appropriate article.

- 5 Various processes are known for processing the reaction mixture in the production of contact lenses, including spincasting and static casting. Spincasting methods are disclosed in U.S. Pat. Nos. 3,408,429 and 3,660,545, and static casting methods are disclosed in U.S. Pat. Nos. 4,113,224 and 4,197,266. The preferred method for producing contact lenses
- 10 comprising the polymer of this invention is by the molding of the silicone hydrogels, which is economical, and enables precise control over the final shape of the hydrated lens. For this method, the reaction mixture is placed in a mold having the shape of the final desired silicone hydrogel, i.e., water-swollen polymer, and the reaction mixture is subjected to conditions whereby
- 15 the monomers polymerize, to thereby produce a polymer/diluent mixture in the shape of the final desired product. Then, this polymer/diluent mixture is treated with a solvent to remove the diluent and ultimately replace it with water, producing a silicone hydrogel having a final size and shape which are quite similar to the size and shape of the original molded polymer/diluent
- 20 article. This method can be used to form contact lenses and is further described in U.S. Pat. Nos. 4,495,313; 4,680,336; 4,889,664; and 5,039,459, incorporated herein by reference.

#### Curing

- 25 Yet another feature of the present invention is a process for curing silicone hydrogel formulations to provide enhanced wettability. It has been found that the gel time for a silicone hydrogel may be used to select cure conditions which provide a wettable ophthalmic device, and specifically a contact lens. The gel time is the time at which a crosslinked polymer network
- 30 is formed, resulting in the viscosity of the curing reaction mixture approaching infinity and the reaction mixture becoming non-fluid. The gel point occurs at a specific degree of conversion, independent of reaction conditions, and therefore can be used as an indicator of the rate of the reaction. It has been found that, for a given reaction mixture, the gel time may be used to

determine cure conditions which impart desirable wettability. Thus, in a process of the present invention, the reaction mixture is cured at or above a gel time that provides improved wettability, or more preferably sufficient wettability for the resulting device to be used without a hydrophilic coating or surface treatment ("minimum gel time"). Preferably improved wettability is a decrease in advancing dynamic contact angle of at least 10% compared to formulation with no high molecular weight polymer. Longer gel times are preferred as they provide improved wettability and increased processing flexibility.

10           Gel times will vary for different silicone hydrogel formulations. Cure conditions also effect gel time. For example the concentration of crosslinker will impact gel time, increasing crosslinker concentrations decreases gel time. Increasing the intensity of the radiation (for photopolymerization) or temperature (for thermal polymerization), the efficiency of initiation (either by selecting a more efficient initiator or irradiation source, or an initiator which absorbs more strongly in the selected irradiation range) will also decrease gel time. Temperature and diluent type and concentration also effect gel time in ways understood by those of skill in the art.

20           The minimum gel time may be determined by selecting a given formulation, varying one of the above factors and measuring the gel time and contact angles. The minimum gel time is the point above which the resulting lens is generally wettable. Below the minimum gel time the lens is generally not wettable. For a contact lens "generally wettable" is a lens which displays an advancing dynamic contact angle of less than about 80°, preferably less than 70° and more preferably less than about 60° or a contact lens which displays a tear film break up time equal to or better than an ACUVUE® lens. Thus, those of skill in the art will appreciate that minimum gel point as defined herein may be a range, taking into consideration statistical experimental variability.

30           In certain embodiments using visible light irradiation minimum gel times of at least about 30 seconds, preferably at least about 35 seconds, and more preferably greater than about 40 seconds have been found to be advantageous.

The mold containing the reaction mixture is exposed to ionizing or actinic radiation, for example electron beams, Xrays, UV or visible light, ie. electromagnetic radiation or particle radiation having a wavelength in the range of from about 150 to about 800 nm. Preferably the radiation source is

5 UV or visible light having a wavelength of about 250 to about 700 nm. Suitable radiation sources include UV lamps, fluorescent lamps, incandescent lamps, mercury vapor lamps, and sunlight. In embodiments where a UV absorbing compound is included in the composition (for example, as a UV block) curing is conducted by means other than UV irradiation (such as by

10 visible light or heat). In a preferred embodiment the radiation source is selected from UVA (about 315 – about 400 nm), UVB (about 280-about 315) or visible light (about 400 –about 450 nm), at low intensity. In another preferred embodiment, the reaction mixture includes a UV absorbing compound, is cured using visible light and low intensity. As used herein the

15 term “low intensity” means those between about 0.1 mW/cm<sup>2</sup> to about 6 mW/cm<sup>2</sup> and preferably between about 0.2 mW/cm<sup>2</sup> and 3 mW/cm<sup>2</sup>. The cure time is long, generally more than about 1 minute and preferably between about 1 and about 60 minutes and still more preferably between about 1 and about 30 minutes. This slow, low intensity cure is critical to providing

20 compatible ophthalmic devices which display lasting resistance to protein deposition *in vivo*.

The temperature at which the reaction mixture is cured is also important. As the temperature is increased above ambient the haze of the resulting polymer decreases. Temperatures effective to reduce haze include

25 temperatures at which the haze for the resulting lens is decreased by at least about 20% as compared to a lens of the same composition made at 25°C. Thus, suitable cure temperatures include those greater than about 25°C, preferably those between about 25°C and 70°C and more preferably those between about 40°C and 70°C. The precise set of cure conditions

30 (temperature, intensity and time) will depend upon the components of lens material selected and, with reference to the teaching herein, are within the skill of one of ordinary skill in the art to determine. Cure may be conducted in one or a multiplicity of cure zones.

The cure conditions must be sufficient to form a polymer network from the reaction mixture. The resulting polymer network is swollen with the diluent and has the form of the mold cavity.

## 5    Deblocking

After the lenses have been cured they are preferably removed from the mold. Unfortunately, the silicone components used in the lens formulation render the finished lenses "sticky" and difficult to release from the lens molds. Lenses can be deblocked (removed from the mold half or tool supporting the  
10    lens) using a solvent, such as an organic solvent. However, in one embodiment of the present invention at least one low molecular weight hydrophilic polymer is added to the reaction mixture, the reaction mixture is formed into the desired article, cured and deblocked in water or an aqueous solution comprising, consisting essentially of and consisting of a small amount  
15    of surfactant. The low molecular weight hydrophilic polymer can be any polymer having a structure as defined for a high molecular weight polymer, but with a molecular weight such that the low molecular weight hydrophilic polymer extracts or leaches from the lens under deblocking conditions to assist in lens release from the mold. Suitable molecular weights include those  
20    less than about 40,000 Daltons, preferably between less than about 20,000 Daltons. Those of skill in the art will appreciate that the foregoing molecular weights are averages, and that some amount of material having a molecular weight higher than the given averages may be suitable, so long as the average molecular weight is within the specified range. Preferably the low  
25    molecular weight polymer is selected from water soluble polyamides, lactams and polyethylene glycols, and mixtures thereof and more preferably polyvinylpyrrolidone, polyethylene glycols, poly 2 ethyl-2-oxazoline (available from Polymer Chemistry Innovations, Tuscon, AZ), poly(methacrylic acid), poly(1-lactic acid), polycaprolactam, polycaprolactone, polycaprolactone diol,  
30    polyvinyl alcohol, poly(2-hydroxyethyl methacrylate), poly(acrylic acid), poly(1-glycerol methacrylate), poly(2-ethyl-2-oxazoline), poly(2-hydroxypropyl methacrylate), poly(2-vinylpyridine N-oxide), polyacrylamide, polymethacrylamide mixtures there of and the like.

The low molecular weight hydrophilic polymer may be used in amounts up to about 20 wt%, more preferably in amounts between about 5 and about 20 wt% based upon the total weight of the reactive components.

Suitable surfactants include non-ionic surfactants including betaines, amine oxides, combinations thereof and the like. Examples of suitable surfactants include TWEEN® (ICI), DOE 120 (Amerchol/Union Carbide) and the like. The surfactants may be used in amounts up to about 10,000, preferably between about 25 and about 1500 ppm and more preferably between about 100 ppm and about 1200 ppm.

Suitable release agents are low molecular weight, and include 1-methyl-4-piperidone, 3-morpholino-1,2-propanediol, tetrahydro-2H-pyran-4-ol, glycerol formal, ethyl-4-oxo-1-piperidine carboxylate, 1,3-dimethyl-3,4,5,6-tetrahydro-2(1H)-pyrimidinone and 1-(2-hydroxyethyl)-2-pyrrolidone.

Lenses made from reaction mixtures without low molecular weight hydrophilic polymer may be deblocked in an aqueous solution comprising at least one organic solvent. Suitable organic solvents are hydrophobic, but miscible with water. Alcohols, ethers and the like are suitable, more specifically primary alcohols and more specifically isopropyl alcohol, DPMA, TPM, DPM, methanol, ethanol, propanol and mixtures thereof being suitable examples.

Suitable deblocking temperatures range from about ambient to about 100°C, preferably between about 70°C and 95°C, with higher temperatures providing quicker deblocking times. Agitation, such as by sonication, may also be used to decrease deblocking times. Other means known in the art, such as vacuum nozzles may also be used to remove the lenses from the molds.

#### Diluent replacement/hydration

Typically after curing the reaction mixture, the resulting polymer is treated with a solvent to remove the diluent (if used), unreacted components, byproducts, and the like and hydrate the polymer to form the hydrogel. Alternatively, depending on the solubility characteristics of the hydrogel's components, the solvent initially used can be an organic liquid such as ethanol, methanol, isopropanol, TPM, DPM, PEGs, PPGs, glycerol, mixtures

thereof, or a mixture of one or more such organic liquids with water, followed by extraction with pure water (or physiological saline). The organic liquid may also be used as a "pre-soak". After demolding (removing the back curve from the lens), lenses may be briefly soaked (times up to about 30 minutes, preferably between about 5 and about 30 minutes) in the organic liquid or a mixture of organic liquid and water. After the pre-soak, the lens may be further hydrated using aqueous extraction solvents.

In some embodiments, the preferred process uses an extraction solvent that is predominately water, preferably greater than 90% water, more preferably greater than 97% water. Other components may include salts such as sodium chloride, sodium borate boric acid, DPM, TPM, ethanol or isopropanol. Lenses are generally released from the molds into this extraction solvent, optionally with stirring or a continuous flow of the extraction solvent over the lenses. This process can be conducted at temperatures from about 2 to about 121°C, preferably from about 20 to about 98°C. The process can be conducted at elevated pressures, particularly when using temperatures in excess of about 100°C, but is more typically conducted at ambient pressures. It is possible to deblock the lenses into one solution (for example containing some release aid) and then transfer them into another (for example the final packing solution), although it may also be possible to deblock the lenses into the same solution in which they are packaged. The treatment of lenses with this extraction solvent may be conducted for a period of from about 30 seconds to about 3 days, preferably between about 5 and about 30 minutes. The selected hydration solution may additionally comprise small amounts of additives such as surfactants. Suitable surfactants include non-ionic surfactants, such as betaines and amine oxides. Specific surfactants include TWEEN 80 (available from Amerchol), DOE 120 (available from Union Carbide), Pluronic, methyl cellulose, mixtures thereof and the like and may be added in amounts between about 0.01 weight % and about 5 weight% based upon total weight of hydration solution used.

In one embodiment the lenses may be hydrated using a "step down" method, where the solvent is replaced in steps over the hydration process. Suitable step down processes have at least two steps, where a percentage of the solvent is replaced with water.

The silicone hydrogels after hydration of the polymers preferably comprise about 10 to about 60 weight percent water, more preferably about 20 to about 55 weight percent water, and most preferably about 25 to about 50 weight percent water of the total weight of the silicone hydrogel. Further details on the methods of producing silicone hydrogel contact lenses are disclosed in U.S. Patents 4,495,313; 4,680,336; 4,889,664; and 5,039,459, which are hereby incorporated by reference.

The cured ophthalmic device of the present invention displays excellent resistance to fouling in vivo, without a coating. When the biomedical device is an ophthalmic device, resistance to biofouling may be measured by measuring the amount of surface deposits on the lens during the wear period, often referred to as "lipid deposits".

Lens surface deposits are measured as follows: Lenses are put on human eyes and evaluated after 30 minutes and one week of wear using a slit lamp. During the evaluation the patient is asked to blink several times and the lenses are manually "pushed" in order to differentiate between deposits and back surface trapped debris. Front and back surface deposits are graded as being discrete (i.e. jelly bumps) or filmy. Front surface deposits give a bright reflection while back surface deposits do not. Deposits are differentiated from back surface trapped debris during a blink or a push-up test. The deposits will move while the back surface trapped debris will remain still. The deposits are graded into five categories based upon the percentage of the lens surface which is effected: none (< about 1%), slight (about 1 to about 5%), mild (about 6% to about 15%), moderate (about 16% to about 25%) and severe (greater than about 25%). A 10% difference between the categories is considered clinically significant.

The ophthalmic devices of the present invention also display low haze, good wettability and modulus.

Haze is measured by placing test lenses in saline in a clear cell above a black background, illuminating from below with a fiber optic lamp at an angle 66° normal to the lens cell, and capturing an image of the lens from above with a video camera. The background-subtracted scattered light image is quantitatively analyzed, by integrating over the central 10 mm of the lens, and

then compared to a -1.00 diopter CSI Thin Lens®, which is arbitrarily set at a haze value of 100, with no lens set as a haze value of 0.

Wettability is measured by measuring the dynamic contact angle or DCA, typically at 23°C, with borate buffered saline, using a Wilhelmy balance.

- 5 The wetting force between the lens surface and borate buffered saline is measured using a Wilhelmy microbalance while the sample is being immersed into or pulled out of the saline. The following equation is used

$$F = 2\gamma p \cos\theta \quad \text{or} \quad \theta = \cos^{-1}(F/2\gamma p)$$

- 10 where F is the wetting force,  $\gamma$  is the surface tension of the probe liquid, p is the perimeter of the sample at the meniscus and  $\theta$  is the contact angle. Typically, two contact angles are obtained from a dynamic wetting experiment – advancing contact angle and receding contact angle. Advancing contact angle is obtained from the portion of the wetting experiment where the sample  
15 is being immersed into the probe liquid, and these are the values reported herein. At least four lenses of each composition are measured and the average is reported.

- However, DCA is not always a good predictor of wettability on eye. The pre-lens tear film non-invasive break-up time (PLTF-NIBUT) is one  
20 measure of in vivo or “clinical” lens wettability. The PLTF-NIBUT is measured using a slit lamp and a circular fluorescent tearscope for noninvasive viewing of the tearfilm (Keeler Tearscope Plus). The time elapsed between the eye opening after a blink and the appearance of the first dark spot within the tear film on the front surface of a contact lens is recorded as PLTF-NIBUT. The  
25 PLTF-NIBUT is measured 30-minutes after the lenses were placed on eye and after one week. Three measurements are taken at each time interval and were averaged into one reading. The PLTF-NIBUT is measured on both eyes, beginning with the right eye and then the left eye.

- Movement is measured using the “push up” test. The patient's eyes  
30 are in the primary gaze position. The push-up test is a gentle digital push of the lens upwards using the lower lid. The resistance of the lens to upward movement is judged and graded according to the following scale: 1 (excessive, unacceptable movement), 2 (moderate, but acceptable

movement), 3 (optimal movement), 4 (minimal, but acceptable movement), 5 (insufficient, unacceptable movement).

The lenses of the present invention display moduli of at least about 30 psi, preferably between about 30 and about 90 psi, and more preferably  
5 between about 40 and about 70 psi. Modulus is measured by using the crosshead of a constant rate of movement type tensile testing machine equipped with a load cell that is lowered to the initial gauge height. A suitable testing machine includes an Instron model 1122. A dog-bone shaped sample having a 0.522 inch length, 0.276 inch "ear" width and 0.213 inch "neck" width  
10 is loaded into the grips and elongated at a constant rate of strain of 2 in/min. until it breaks. The initial gauge length of the sample ( $L_0$ ) and sample length at break ( $L_f$ ) are measured. Twelve specimens of each composition are measured and the average is reported. Tensile modulus is measured at the initial linear portion of the stress/strain curve.

15 The contact lenses prepared by this invention have  $O_2$  Dk values between about 40 and about 300 barrer, determined by the polarographic method. Lenses are positioned on the sensor then covered on the upper side with a mesh support. The lens is exposed to an atmosphere of humidified 2.1%  $O_2$ . The oxygen that diffuses through the lens is measured using a  
20 polarographic oxygen sensor consisting of a 4 mm diameter gold cathode and a silver ring anode. The reference values are those measured on commercially available contact lenses using this method. Balafilcon A lenses available from Bausch & Lomb give a measurement of approx. 79 barrer. Etafilcon lenses give a measurement of 20 to 25 barrer. ( $1 \text{ barrer} = 10^{-10} (\text{cm}^3 \text{ of gas} \times \text{cm}^2)/(\text{cm}^3 \text{ of polymer} \times \text{sec} \times \text{cm Hg})$ ).  
25

Gel time is measured using the following method. The photo-polymerization reaction is monitored with an ATS StressTech rheometer equipped with a photo-curing accessory, which consists of a temperature-controlled cell with a quartz lower plate and an aluminum upper plate, and a  
30 radiation delivery system equipped with a bandpass filter. The radiation, which originates at a Novacure mercury arc lamp equipped with an iris and computer-controlled shutter, is delivered to the quartz plate in the rheometer via a liquid light guide. The filter is a 420 nm (20 nm FWHM) bandpass filter, which simulates the light emitted from a TL03 bulb. The intensity of the

radiation, measured at the surface of the quartz window with an IL1400A radiometer, using an XRL140A sensor, is controlled to  $\pm 0.02$  mW/cm<sup>2</sup> with an iris. The temperature is controlled at  $45 \pm 0.1^\circ\text{C}$ . After approximately 1 mL of the de-gassed reactive mixture is placed on the lower plate of the

5 rheometer, the 25 mm diameter upper plate is lowered to  $0.500 \pm 0.001$  mm above the lower plate, where it is held until after the reaction reached the gel point. The sample is allowed to reach thermal equilibrium (~4 minutes, determined by the leveling-off of the steady shear viscosity) before the lamp shutter is opened and the reaction begun. During this time while the sample

10 is reaching thermal equilibrium, the sample chamber is purged with nitrogen gas at a rate of 400 sccm. During the reaction the rheometer continuously monitors the strain resulting from an applied dynamic stress (fast oscillation mode), where time segments of less than a complete cycle are used to calculate the strain at the applied programmable stress. The computer

15 calculates the dynamic shear modulus ( $G'$ ), loss modulus ( $G''$ ), and viscosity ( $\eta^*$ ), as a function of exposure time. As the reaction proceeds the shear modulus increases from  $<1$  Pa to  $>0.1$  MPa, and  $\tan \delta (=G''/G')$  drops from near infinity to less than 1. For measurements made herein the gel time is the time at which  $\tan \delta$  equals 1 (the crossover point when  $G'=G''$ ). At the time

20 that  $G'$  reaches 100 Pa (shortly after the gel point), the restriction on the upper plate is removed so that the gap between the upper and lower plates can change as the reactive monomer mix shrinks during cure.

It will be appreciated that all of the tests specified above have a certain amount of inherent test error. Accordingly, results reported herein are not to

25 be taken as absolute numbers, but numerical ranges based upon the precision of the particular test.

In order to illustrate the invention the following examples are included. These examples do not limit the invention. They are meant only to suggest a method of practicing the invention. Those knowledgeable in contact lenses as

30 well as other specialties may find other methods of practicing the invention. However, those methods are deemed to be within the scope of this invention.

## EXAMPLES

The following abbreviations are used in the examples below:

	SiGMA	2-propenoic acid, 2-methyl-,2-hydroxy-3-[3-[1,3,3,3-tetramethyl-1-[(trimethylsilyl)oxy]disiloxanyl]propoxy]propyl ester
	DMA	N,N-dimethylacrylamide
5	HEMA	2-hydroxyethyl methacrylate
	mPDMS	800-1000 MW (M <sub>n</sub> ) monomethacryloxypropyl terminated mono-n-butyl terminated polydimethylsiloxane
	Norbloc	2-(2'-hydroxy-5-methacryloxyethylphenyl)-2H-benzotriazole
10	CGI 1850	1:1 (wgt) blend of 1-hydroxycyclohexyl phenyl ketone and bis(2,6-dimethoxybenzoyl)-2,4,4-trimethylpentyl phosphine oxide
	PVP	poly(N-vinyl pyrrolidone) (K value 90)
	Blue HEMA	the reaction product of Reactive Blue 4 and HEMA, as described in Example 4 of U.S. Pat. no. 5,944,853
15	IPA	isopropyl alcohol
	D3O	3,7-dimethyl-3-octanol
	mPDMS-OH	mono-(3-methacryloxy-2-hydroxypropyloxy)propyl terminated, mono-butyl terminated polydimethylsiloxane (MW 1100)
	TEGDMA	tetraethyleneglycol dimethacrylate
20	TrEGDMA	triethyleneglycol dimethacrylate
	TRIS	3-methacryloxypropyltris(trimethylsiloxy)silane
	MPD	3-methacryloxypropyl(pentamethyldisiloxane)
	MBM	3-methacryloxypropylbis(trimethylsiloxy)methylsilane
	AcPDMS	bis-3-methacryloxy-2-hydroxypropyloxypropyl polydimethylsiloxane
25	Triglide	tripropyleneglycol methyl ether
	CGI 819	bis(2,4,6-trimethylbenzoyl)-phenylphosphineoxide
	PVP low MW	poly(N-vinylpyrrolidone) (K value 12)

30 Throughout the Examples intensity is measured using an IL 1400A radiometer, using an XRL 140A sensor.

#### Examples 1-10

The reaction components and diluent (D30) listed in Table 1 were mixed together with stirring or rolling for at least about 3 hours at about 23°C, until all components were dissolved. The reactive components are reported as weight percent of all reactive components and the diluent is weight percent of final reaction mixture. The reaction mixture was placed into thermoplastic contact lens molds (made from Topas® copolymers of ethylene and norbornene obtained from Ticona Polymers), and irradiated using Philips TL 20W/03T fluorescent bulbs at 45°C for about 20 minutes in N<sub>2</sub>. The molds were opened and lenses were extracted into a 50:50 (wt) solution of IPA and H<sub>2</sub>O, and soaked in IPA at ambient temperature for about 15 hours to remove residual diluent and monomers, placed into deionized H<sub>2</sub>O for about 30 minutes, then equilibrated in borate buffered saline for at least about 24 hours and autoclaved at 122°C for 30 minutes. The properties of the resulting lenses are shown in Table 1.

Table 1

EX. #	1	2	3	4	5	6	7	8	9	10
Comp.										
SiGMA	28	30	28.6	28	31	32	29	39.4	20	68
PVP (K90)	7	10	7.1	7	7	7	6	6.7	3	7
DMA	23.5	17	24.5	23.5	20	20	24	16.4	37	22
MPDMS	31	32	0	31	31	34	31	29.8	15	0
TRIS	0	0	0	0	0	0	0	0	15	0
HEMA	6	6	6.1	6	6.5	3	5.5	2.9	8	0
Norbloc	2	2	0	2.0	2	2	2	1.9	0	0
CGI 1850	0.98	1	1.02	1	1	1	1	1	1	0
TEGDMA	1.5	2	1.02	1.5	1.5	1	1.5	1.9	0	2
TrEGDMA	0	0	0	0	0	0	0	0	1	0
Blue HEMA	0.02	0	0	0	0	0	0	0	0	0
mPDMS-OH	0	0	31.6	0	0	0	0	0	0	0
Darocur 1173	0	0	0	0	0	0	0	0	0	1
D30 %	23	26	17	23	23	29	32	28	17	27
<b>Properties</b>										
%EWC <sup>1</sup>	36	33	39	40	36	37	39	25	48	29

Modulus (psi)	68	78	112	61	67	50	66	92	43	173
% Elongation	301	250	147	294	281	308	245	258	364	283
DCA <sup>2</sup> (advancing)	62	55	58	64	72	65	61	55	92	72
Dk <sup>3</sup> (edge corrected)	103	111	101	131	110	132	106	140	64	76

1. Equilibrium water content
2. Dynamic contact angle, measured with physiological borate-buffered saline using a Wilhelmy balance.
3. Oxygen permeability, edge corrected, in Barrers.

5

The results of Examples 1-10 show that the reaction mixture components and their amounts may be varied substantially, while still providing uncoated lenses having an excellent balance of mechanical properties and wettability. The contact angle (DCA) of Example 9 may be too high to form a lens that would be clinically wettable, and the modulus may be lower than desired to provide a mechanically robust lens. Example 9 contained the lowest concentration of SiGMA (20%). Because the SiGMA had been reduced, less PVP could be added to the formulation and still provide a compatible reaction mixture. Thus, these examples show that SiGMA is effective in compatibilizing PVP and that when sufficient SiGMA and PVP are present lenses with desirable wettability and other mechanical properties can be made without any form of surface modification.

#### Example 11

Lenses having the formulation of Example 1 were remade, without controlling cure intensity. The mechanical properties are reported in Table 2, below. These lenses were clinically evaluated using ACUVUE® 2 lenses as controls. The lenses were worn by 6 patients in a daily wear mode (nightly removal) for a period of one week. At one week the PLTF-NIBUT was 3.6 (±3.0) seconds compared to 5.8 (±2.5) seconds for ACUVUE® 2 lenses. The front surface deposition was graded none to slight for 50% of the test lenses and 100% for the control lenses. The movement was acceptable for both test and control lenses.

**Example 12**

Example 11 was repeated except that the cure intensity was reduced to 1.0 mW/cm<sup>2</sup>. The mechanical properties are reported in Table 2, below. These lenses were clinically evaluated using ACUVUE® 2 lenses as controls.

- 5 The test lenses were worn by 15 patients in a daily wear mode (nightly removal), in one eye for a period of one week and an ACUVUE® 2 lens was worn in the contralateral eye. At one week the PLTF-NIBUT was 8.2 (±1.7) seconds compared to 6.9 (±1.5) seconds for ACUVUE® 2 lenses. The front surface deposition was graded none to slight for all of the patients for both
- 10 test and control lenses. The movement was acceptable for both test and control lenses.

**Table 2**

Ex.#	1	11	12
%EWC	36	36	36
Modulus (psi)	68	74	87
Elongation	301	315	223
DCA	62	77	56
Dk	103	127	102

- Generally the mechanical properties for Examples 1, 11 and 12 are
- 15 consistent results for multiple runs of the same material. However, the clinical results for Examples 11 (uncontrolled cure intensity) and 12 (low, controlled cure intensity) are substantially different. The on eye wettability after one week of wear for Example 11 (measured by PLTF-NIBUT) was worse than the ACUVUE® 2 lenses (3.6 v. 5.8) and half the lenses had more than slight
- 20 surface depositions. The Example 12 lenses (controlled, low intensity cure) displayed significantly improved on-eye wettability, which was measurably better than ACUVUE® 2 lenses (8.2 v. 6.9) and no surface depositions. Thus, using a low, controlled cure provides an uncoated lens having on-eye wettability which is as good as, and in some cases better than conventional
- 25 hydrogel lenses.

**Examples 13-17**

- Reaction mixtures described in Table 3 and containing low or no hydroxyl-functionalized silicone containing monomer (in these Examples 5 SiGMA) were mixed with constant stirring at room temperature for 16 hours. Even after 16 hours each of the reaction mixtures remained cloudy and some contained precipitates. Accordingly, these reaction mixtures could not be used to produce lenses.

**Table 3**

Ex. #	13	14	15	16	17
Composition					
SiGMA	0	0	0	10	20
PVP (K90)	12	12	10	8.0	8.0
DMA	10	10	8.3	19	19
MPDMS	37	37	30.8	35	28
TRIS	14	14	11.7	17	14
HEMA	25	25	37.5	8.0	8.0
TEGDMA	1.0	1.0	0.83	2.0	2.0
Darocur 1173	1.0	1.0	0.83	1.0	1.0
D30 %	23	31	31	27	27

10

- Examples 13 through 15 show that reaction mixtures without any hydroxyl-functionalized silicone containing monomer (SiGMA or mPDMS-OH) are incompatible, and not suitable for making contact lenses. Examples 16 and 17 show that concentrations of hydroxyl-functionalized silicone containing monomer less than about 20 weight% are insufficient to compatibilize significant amounts of high molecular weight PVP. However, comparing Example 17 to Example 9, lesser amounts of high molecular weight PVP (3 weight %) can be included and still form a compatible reaction mixture.

20

Examples 18-25

A solution of 1.00 gram of D3O, 1.00 gram of mPDMS and 1.00 gram of TRIS was placed in a glass vial (Ex. 19). As the blend was rapidly stirred at about 20 to 23 °C with a magnetic stir bar, a solution of 12 parts (wt) PVP (K90) and 60 parts DMA was added dropwise until the solution remained cloudy after 3 minutes of stirring. The mass of the added DMA/PVP blend was determined in grams and reported as the "monomer compatibility index". This test was repeated using SiGMA (Ex. 18), MBM (Ex. 20), MPD (Ex. 21), acPDMS, where n=10 (Ex. 22), acPDMS where n=20 (Ex. 23), iSiGMA-3Me (Ex. 24) and TRIS2-HOEOP2 (Ex. 25) as test silicone monomers in place of TRIS.

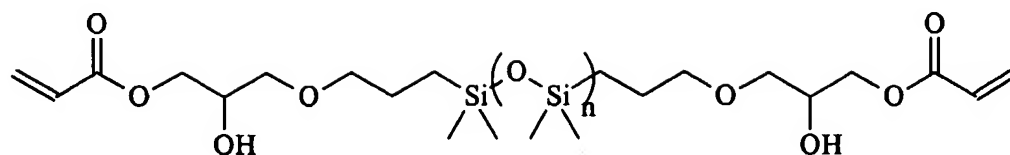
Table 4

Ex. #	Test silicone-containing monomer	Monomer compatibility index	Si:OH
18	SiGMA	1.8	3:1
19	TRIS	0.07	4:0
20	MBM	0.09	3:0
21	MPD	0.05	2:0
22	acPDMS (n=10)*	1.9	11:2
23	acPDMS (n=20)*	1	21:2
24	iSiMAA-3Me	0.15	4:0
25	TRIS2-HOEOP2	0.11	3:2
26	MPDMS-OH	0.64	~11:1

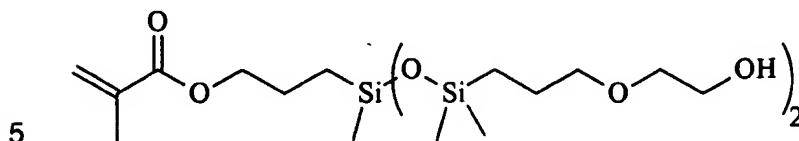
15

Structures for acPDMS, iSiGMA-3Me and TRIS2-HOEOP2 are shown below.

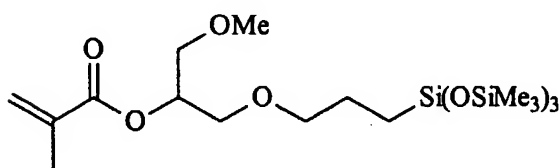
acPDMS (n averages 10 or 20):



TRIS2-HOEOP2



iSiMAA3-Me



The results, shown in Table 4, show that SiGMA, acPDMS (where  $n = 10$  and 20) and mPDMS-OH more readily incorporate into a blend of a diluent, another silicone containing monomer, a hydrophilic monomer, and an high molecular weight polymer (PVP) than alternative silicone-containing monomers. Thus, compatibilizing silicone containing monomers having a compatibility index of greater than about 0.5 are useful for compatibilizing high molecular weight hydrophilic polymers like PVP.

#### Example 27 -35

Lenses were made using the reaction mixture formulation of Example 1. The plastic contact lens molds (made from Topas® copolymers of ethylene and norbornene obtained from Ticona Polymers) were stored overnight in nitrogen ( $<0.5\% \text{ O}_2$ ) before use. Each mold was dosed with 75  $\mu\text{l}$  reaction mixture. Molds were closed and lenses photocured using the times and cure intensities indicated in Table 5. Lenses were formed by irradiation of the monomer mix using visible light fluorescent bulbs, curing at  $45^\circ\text{C}$ . The intensity was varied by using a variable balast or light filters, in two steps of varied intensity and cure time. The step 2 time was selected to provide the same total irradiation energy (about  $830 \text{ mJ/cm}^2$ ) for each sample.

The finished lenses were demolded using a 60:40 mixture of IPA/ water. The lenses were transferred to a jar containing 300 g 100% isopropyl alcohol (IPA). The IPA was replaced every 2 hours for 10 hours. At the end of about 10 hours, 50% of the IPA was removed and replaced with DI water and the jar was rolled for 20 minutes. After 20 minutes, 50% of the IPA was removed and replaced with DI water and the jar was rolled for another 20 minutes. The lenses were transferred to packing solution, rolled for 20 minutes and then tested.

10

**Table 5**

Ex.#	Step 1 intensity (mW/cm <sup>2</sup> )	Step 1 time (min:sec)	Step 2 intensity (mW/cm <sup>2</sup> )	Step 2 time (min:sec)	Advancing Contact Angle
27	1.1	6:55	5.5	1:28	51 ± 1
28	1.1	2:46	5.5	2:21	55 ± 2
29	1.1	11:03	5.5	0:35	55 ± 1
30	1.7	6:30	5.5	0:35	50 ± 1
31	1.7	1:37	5.5	2:21	55 ± 1
32	1.7	4:04	5.5	1:28	54 ± 2
33	2.4	2:52	5.5	1:28	62 ± 6
34	2.4	4:36	5.5	0:35	76 ± 9
35	2.4	1:09	5.5	0:35	78 ± 6

The contact angles for Examples 27 through 32 are not significantly different, indicating that step 1 cure intensities of less than about 2 mW/cm<sup>2</sup> provide improved wettability for this lens formulation regardless of the step 1 cure time. However, those of skill in the art will appreciate that shorter step 1 cure times (such as those used in Examples 28 and 31) allow for shorter overall cure cycles. Moreover, it should be noted that even though the contact angles for Examples 33 through 35 are measurably higher than those of Examples 27-32, the lenses of Examples 33-35 may still provide desirable on eye wettability.

#### Examples 36 - 41

The reaction components of Example 1, were blended with either 25% or 40% D3O as diluent in accordance with the procedure of Example 1. The resultant reaction mixtures were charged into plastic contact lens molds (made from Topas® copolymers of ethylene and norbornene obtained from

Ticona Polymers) and cured in a glove box under a nitrogen atmosphere, at about 2.5 mW/cm<sup>2</sup> intensity, about 30 minutes and the temperatures shown in Table 6, below. The lenses were removed from the molds, hydrated and autoclaved as describe in Example 1. After hydration the haze values of the lenses were determined. The results shown in Table 6 show that the degree of haziness was reduced at the higher temperatures. The results also show that as the concentration of diluent decreases the haze also decreases.

Table 6

Ex. #	%D30	Temp. (°C)	% haze	DCA(°)
36	25	25	30 (6)	99
37	25	50-55	12 (2)	100
38	25	60-65	14 (0.2)	59
39	40	25	50 (10)	68
40	40	50-55	40 (9)	72
41	40	60-65	32 (1)	66

Haze (std. dev.)

The results in Table 6 show that haze may be reduced by about 20% (Example 41 v. Example 39) and up to as much as about 65% (Example 37 v. Example 36) by increasing the cure temperature. Decreasing diluent concentration from 40 to 25% decrease haze by between about 40 and 75%.

#### Examples 42-47

Lenses were made from the formulations shown in Table 7 using the procedure of Example 1, with a 30 minute cure time at 25°C and an intensity of about 2.5 mW/cm<sup>2</sup>. Percent haze was measured and is reported in Table 8.

Table 7

Ex.#	42	43	44	45	46	47
SiGMA	28.0	28.0	28.0	28.0	28.0	28.0
mPDMS	31.0	31.0	28.0	28.0	28.0	28.0
acPDMS (n=10)	0.0	0.0	4.0	4.0	4.0	4.0
DMA	23.5	23.5	23.5	23.5	24.0	24.0
HEMA	6.0	6.0	5.0	5.0	6.0	6.0
TEGDMA	1.5	1.5	1.5	1.5	0.0	0.0
Norbloc	2.0	2.0	2.0	2.0	2.0	2.0

PVP (K-90)	7.0	7.0	7.0	7.0	7.0	7.0
CGI 1850	1.0	1.0	1.0	1.0	1.0	1.0
D30	25.0	40	25.0	40.0	25.0	40.0
<b>Properties</b>						
Haze	30	50	7.3	14	26	25
Modulus (psi)	74	56	148	104	74	NT
Elongation (%)	326	395	188	251	312	NT
EWC(%)	38	41	33	35	38	39

A comparison of the results for formulations having the same amount of diluent and either TEGDMA or acPDMS (Examples 42 and 46 and Examples 43 and 47) shows that acPDMS is an effective crosslinker and provides lenses with properties which are comparable to those where TEGDMA is used as a crosslinker. Examples 44 and 45 contain both crosslinkers. Haze for these Examples decreased substantially compared to the lenses made from either crosslinker alone. However, modulus and elongation were negatively impacted (likely because the amount of crosslinker was too great).

#### Examples 48-54

Reaction mixtures were made using the formulations shown in Table 8 with the diluents indicated. The reaction mixtures were placed into thermoplastic contact lens molds, and irradiated using Philips TL 20W/03T fluorescent bulbs at 45°C, 0.8 mW/cm<sup>2</sup> for about 32 minutes. The molds were opened and lenses were released into deionized water at 95°C over a period of 20 minutes. The lenses were then placed into borate buffered saline solution for 60 minutes and autoclaved at 122°C and 30 minutes. The properties of the resulting lenses are shown in Table 9.

**Table 8**

Ex. #	48	49	50	51	52	53	54
Components							

SIGMA	30	30	30	33	34	25	20
PVP	6	6	6	6	7	6	6
DMA	31	31	31	30	30	31	31
MPDMS	19	22	23.5	16.5	19	25	28
AcPDMS (n=10)	2	0	0	3	0	0	0
HEMA	9.85	8.5	6.95	9	6	10.5	12.5
Norbloc	1.5	1.5	1.5	2	1.5	1.5	1.5
CGI 819	0.23	0.23	0.25	0.48	0	0.23	0.23
CGI 1850	0	0	0	0	1	0	0
EGDMA	0.4	0.75	0.8	0	0	0.75	0.75
TEGDMA	0	0	0	0	1.5	0	0
Blue HEMA	0.02	0.02	0	0	0	0.02	0.02
% Diluent	40	40	27.3	39.4	25.9	40	40
Diluent comp	A	A	B	C	D	A	A
<b>Properties</b>							
EWC (%)	45	45	47	49	47	49	50
DCA (advancing)	52(2)	51(7)	74(10)	108	75(6)	47(2)	56(11)
Modulus (psi)	91	77	69	55	49	63	67
Elongation (%)	NT	232	167	275	254	110	124
Dk (barrers)	54	60	78	44	87	59	60

Diluents (weight parts):

A = 72.5% t-amyl alcohol and 27.5 PVP ( $M_w = 2500$ )

B = t-amyl alcohol

C = 15/38/38% TMP/2M2P/ PVP ( $M_w = 2500$ )

5 D= 57/43 2M2P/TMP

NT – not tested

Thus, Examples 48 and 51 show that formulations comprising both  
 10 hydrophilic (EGDMA or TEGDMA) and hydrophobic crosslinkers (acPDMS)  
 provide silicone hydrogel compositions which display an excellent balance of  
 properties including good water content, moderate Dk, wettability, modulus  
 and elongation.

#### 15 Example 55

The lenses of Example 48 were clinically evaluated. The lenses were  
 worn by 18 patients in a daily wear mode (nightly removal) for a period of one  
 week. At one week the PLTF-NIBUT was 8.4 ( $\pm 2.9$ ) seconds compared to 7.0

( $\pm 1.3$ ) seconds for ACUVUE® 2 lenses. The front surface discrete deposition was graded none to slight for 97% of the patients with the test lenses, compared with 89% in control lenses. The movement was acceptable for both test and control lenses.

5

#### Example 56

The lenses of Example 49 were clinically evaluated. The lenses were worn by 18 patients in a daily wear mode (nightly removal) for a period of one week. At one week the PLTF-NIBUT was 8.4 ( $\pm 2.9$ ) seconds compared to 7  
10 ( $\pm 1.3$ ) seconds for ACUVUE® 2 lenses. The front surface discrete deposition was graded none to slight for 95% of the patients with the test lenses, compared with 89% in control lenses. The movement was acceptable for both test and control lenses.

#### 15 Example 57

The lenses of Example 51 were clinically evaluated. The lenses were worn by 13 patients in a daily wear mode (nightly removal) for a period of one week. At one week the PLTF-NIBUT was 4.3 ( $\pm 1.9$ ) seconds compared to 9.6  
20 ( $\pm 2.1$ ) seconds for ACUVUE® 2 lenses. The front surface discrete deposition was graded none to slight for 70% of the patients with the test lenses, compared with 92% in control lenses. The movement was acceptable for both test and control lenses. Thus, there is some correlation between contact angle measurements (108° for Example 51 versus 52° for Example 48) and clinical wettability as measure by PLTF-NIBUT (4.3 seconds for Example 51  
25 versus 8.4 seconds for Example 48).

#### Examples 58-60

Silicone hydrogel lenses were made using the components (expressed  
30 in weight parts) listed in Table 9 and the following procedure:

The components were mixed together in a jar to for a reaction mixture. The jar containing the reaction mixture was placed on a jar mill roller and rolled overnight.

The reaction mixture was placed in a vacuum desiccator and the  
35 oxygen removed by applying vacuum for 40 minutes. The desiccator was

- back filled with nitrogen. Contact lenses were formed by adding approximately 0.10 g of the degassed lens material to the concave front curve side of TOPAS® (copolymers of ethylene and norbornene obtained from Ticona Polymers) mold cavities in a glove box with nitrogen purge. The molds
- 5 were closed with polypropylene convex base curve mold halves.
- Polymerization was carried out under a nitrogen purge and was photoinitiated with 5 mW cm<sup>2</sup> of visible light generated using 20W fluorescent lights with a TL-03 phosphor. After curing for 25 minutes at 45°C, the molds were opened.
- 10 The concave front curve portion of the lens mold was placed into a sonication bath (Aquasonic model 75D) containing deionized water under the conditions (temperature and amount of Tween) shown in Table 10. The lens deblock time is shown in Table 10. The lenses were clear and of the proper shape to be contact lenses.

Table 9

	Ex. 58	Ex. 59	Ex. 60	Ex. 61
SIGMA	3.05	3.2	3.2	3.0
mPDMS	1.7	1.7	1.7	1.7
DMA	3.2	3.0	3.1	3.2
PVP	0.6	0.6	0.6	0.6
HEMA	1.0	0.8	0.8	1.0
TEGDMA	0.2	0.4	0.3	0.2
Norblock	0.15	0.2	0.2	0.2
CGI 1850	0.1	0.1	0.3	0.3
Triglide	1.5	1.5		1.5
2M2P	2.5	2.5	2.5	2.5
PVP low MW	0.5	1.5	1.5	0.5

15

Table 10

Ex. #	Form. Ex. #	[Tween] (ppm)	Temp (°C)	Deblock time (min.)
62	58	850	75	10
63	58	10,000	70	10-15
64	58	0	75	20-22
65	58	850	22	10-15
66	59	850	85	3
67	60	850	85	6
68	61	850	75	18

Example 69

The lenses of Example 59 which were deblocked in Example 66, were further hydrated in deionized water at 65°C for 20 minutes. The lenses were then transferred into borate buffered saline solution and allowed to equilibrate for at least about 24 hours. The lenses were clear and of the proper shape to be contact lenses. The lenses had a water content of 43%, a modulus of 87 psi, an elongation of 175%, and a Dk of 61 barriers. The lenses were found to have an advancing contact angle of 57 degrees. This indicates the lens surfaces were substantially free of leachable hydrophobic material.

Example 70

The concave front curve portion of the lens mold from Example 61 was placed into a sonication bath (Aquasonic model 75D) containing about 5% DOE-120 in deionized water at about 75°C. The lenses deblocked from the frame in 18 minutes.

Example 71 (use of an organic solvent)

The concave front curve portion of the lens mold from example 61 was placed into a sonication bath (Aquasonic 75D) containing about 10% of 2-propanol in deionized water at 75°C. The lenses deblocked from the frame in 15 minutes. When Tween was used as the additive (Example 68) the deblock time was 18 minutes. Thus, the present example shows that organic solvents may also be used to deblock lenses comprising low molecular weight hydrophilic polymers.

EXAMPLE 72 (contains no low molecular weight PVP)

Silicone hydrogel lenses were made using the formulation and procedure of Example 58, but without any low molecular weight PVP. The following procedure was used to deblock the lenses.

The concave front curve portion of the lens mold was placed into a sonication bath (Aquasonic model 75D) containing about 850ppm of Tween in deionized water at about 65°C. The lenses did not release from the mold. The deblock time for the formulation which contained low molecular weight hydrophilic polymer (Example 58 formulation) under similar deblock conditions

(Example 62 - 850 ppm Tween and 75°C) was 10 minutes. Thus, the present Example shows that deblocking cannot be accomplished in water only, in this formulation without including low molecular weight hydrophilic polymer in the formulation.

5

### Example 73

The concave front curve portion of the lens mold from example 72 was placed into a sonication bath (Aquasonic 75D) containing about 10% of 2-propanol an organic solvent in deionized water at 75°C. The lenses  
10 deblocked from the frame in 20 to 25 minutes. Thus, lenses of the present invention which do not contain low molecular weight hydrophilic polymer may be deblocked using an aqueous solution comprising an organic solvent.

### Examples 74-76

15 Formulations and lenses were made according to Example 49, but with varying amounts of photoinitiator (0.23, 0.38 or 0.5 wt.%), curing at 45°C with Philips TL 20W/03T fluorescent bulbs (which closely match the spectral output of the visible light used to measure gel time) irradiating the molds at 2.0 mW/cm<sup>2</sup>. The advancing contact angles of the resulting lenses are shown in  
20 Table 11.

Table 11

Ex. #	Wt%	Advancing DCA	Gel time (sec)
74	0.23	59 (4)	65
75	0.38	62 (6)	57
76	0.5	80 (7)	51

25

### Examples 77-79

Gel times were measured for the formulation of Example 1 at 45°C at 1.0, 2.5 and 5.0 mW/cm<sup>2</sup>. The results are shown in Table 12.

30

Table 12

Ex. #	Intensity (mW/cm <sup>2</sup> )	gel time (sec)
77	1	52

78	2.5	38
79	5	34

The results of Examples 74 through 76 and 77 through 79 compared with Examples 27-35, show that as gel times increase, wettability improves.

- 5 Thus, gel points can be used, in coordination with contact angle measurements, to determine suitable cure conditions for a given polymer formulation and photoinitiator system.

#### Examples 79-83

- 10 Reaction mixtures were made using reactive components shown in Table 14 and 29% (based upon all reactive components and diluent) t-amyl alcohol as a diluent and 11% PVP 2,500 (based upon reactive components). Amounts indicated are based upon 100% reactive components. The reaction mixtures were placed into thermoplastic contact lens molds, and irradiated
- 15 using Philips TL 20W/03T fluorescent bulb at 60°C, 0.8 mW/cm<sup>2</sup> for about 30 minutes under nitrogen. The molds were opened and lenses were released into deionized water at 95°C over a period of 15 minutes. The lenses were then placed into borate buffered saline solution for 60 minutes and autoclaved at 122° C for 30 min. The properties of the resulting lenses are shown in
- 20 Table 13.

**Table 13**

Ex. #	79	80	81	82	83
<b>Components</b>					
SiGMA	30	30	30	30	30
PVP	0	1	3	6	8
DMA	37	36	34	31	29
MPDMS	22	22	22	22	22
HEMA	8.5	8.5	8.5	8.5	8.5
Norbloc	1.5	1.5	1.5	1.5	1.5
CGI 819	0.25	0.25	0.25	0.25	0.25
EGDMA	0.75	0.75	0.75	0.75	0.75
<b>Properties</b>					
DCA (advancing)	122(8)	112(6)	66(13)	58(8)	54(3)
Haze	18(4)	11(1)	13(1)	14(2)	12(1)

Table 12 shows that the addition of PVP dramatically decreases contact angle. As little as 1% decreases the dynamic contact angle by about 10% and as little as 3% decreases dynamic contact angle by about 50%.

5

#### Examples 84-86

Silicone hydrogel lenses were made using the components listed in Table 14 and the following procedure:

The reactive components and diluent were mixed together at room temperature in a jar. The jar containing the reaction mixture was placed on a jar mill roller and rolled overnight.

The reaction mixture was placed in a vacuum desiccator and the oxygen removed by applying vacuum for 40 minutes. The desiccator was back filled with nitrogen. Contact lenses were formed by adding approximately 0.10 g of the degassed lens material to the concave front curve side of TOPAS® mold cavities in a glove box with nitrogen purge. The molds were closed with polypropylene convex base curve mold halves. Polymerization was carried out under a nitrogen purge and was photoinitiated with 1.0 mW cm<sup>2</sup> of visible light generated using 20W fluorescent lights with a TL-03 phosphor. After curing for 15 minutes at 45°C, the molds were opened. The concave front curve portion of the lens mold was placed into de-ionized water at 95-100 °C. The lens deblock time is shown in Table 14. The lenses were clear and of the proper shape to be contact lenses.

25

**Table 14**

Ex. #	84	85	86
Components			
SiGMA	30	30	30
PVP	6	6	6
DMA	31	31	31
MPDMS	21	21	21
AcPDMS (n=10)	0	0	0
HEMA	9.25	9.25	9.25
Norbloc	1.5	1.5	1.5
CGI 819	0.25	0.25	0.25
CGI 1850	0	0	0

EGDMA	1.0	1.0	1.0
TEGDMA	0	0	0
Blue HEMA	0	0	0
% Diluent	40.0	40.0	40.0
Diluent comp	E	F	G

DCA (advancing)	53(3)	55(3)	56(9)
Haze	15(3)	22(2)	16(2)
Deblock Time (min)	3	5	8

Diluents (weight parts of diluent):

E = 62.5% t-amyl alcohol and 37.5% PVP ( $M_w = 2500$ )

F = 62.5% t-amyl alcohol and 37.5% 1,3-dimethyl-3,4,5,6-tetrahydro-2(1H)-pyrimidinone

G = 62.5% t-amyl alcohol and 37.5% ethyl-4-oxo-1-piperidinecarboxylate

Thus, Examples 84-86 show that a variety of release agents are useful for improving deblocking.

#### 10 Example 87

A reaction mixture was made using reactive components shown in Table 15 and 42% (based upon all reactive components and diluent) t-amyl alcohol as a diluent. The reaction mixtures were placed into thermoplastic contact lens molds, and irradiated using Philips TL 20W/03T fluorescent bulb at 50°C, 0.8 mW/cm<sup>2</sup> for about 30 minutes under nitrogen. The molds were opened and lenses were released into deionized water at room temperature over a period of 15 minutes. The lenses were then placed into borate buffered saline solution for 60 minutes and autoclaved at 122° C for 30 min. The properties of the resulting lenses are shown in Table 12.

**Table 12**

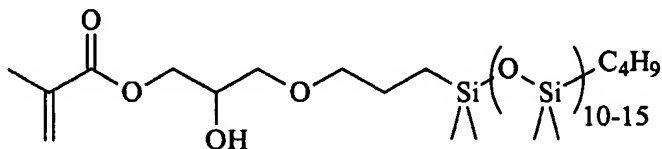
Ex. #	87
Components	
SIGMA	0
PVP	7
DMA	25
MPDMS	48.6
AcPDMS	5
HEMA	12.25

Norbloc	1.5
CGI 819	0.25
TEGDMA	0.4

**Example 88**

Preparation of mPDMS-OH (used in Examples 3)

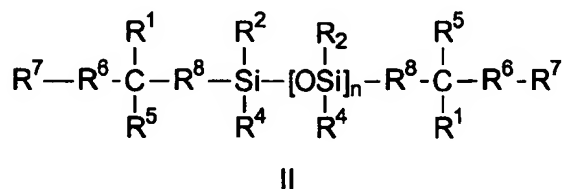
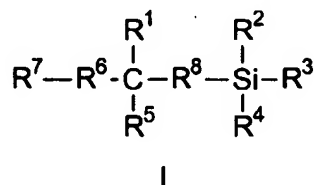
- 5        96 g of Gelest MCR-E11 (mono-(2,3-epoxypropyl)-propyl ether terminated polydimethylsiloxane(1000 MW)), 11.6 g methacrylic acid, 0.10 g triethylamine and 0.02 g hydroquinone monomethylether were combined and heated to 140°C with an air bubbler and with stirring for 2.5 hours. The product was extracted with saturated aqueous NaHCO<sub>3</sub> and CH<sub>2</sub>Cl<sub>2</sub>. The
- 10    CH<sub>2</sub>Cl<sub>2</sub> layer was dried over Na<sub>2</sub> SO<sub>4</sub> and evaporated to give 94 g of product. HPLC/MS was consistent with desired structure:



15

What is claimed is

1. A biomedical device formed from a reaction mixture comprising a high molecular weight hydrophilic polymer and an effective amount of an hydroxyl-functionalized silicone-containing monomer.
2. The biomedical device of claim 1 wherein the effective amount of said hydroxyl-functionalized silicone-containing monomer is about 5% to about 90%
3. The biomedical device of claim 1 wherein the device is a silicone hydrogel contact lens.
4. The biomedical device of claim 1 comprising about 1% to about 15% high molecular weight hydrophilic polymer.
5. The biomedical device of claim 2 wherein the effective amount of hydroxyl-functionalized silicone-containing monomer is about 10% to about 80%.
6. The biomedical device of claim 1 wherein said hydroxyl-functionalized silicone-containing monomer is a compound of Formula I or II



wherein:

n is an integer between 3 and 35

R<sup>1</sup> is hydrogen, C<sub>1-6</sub>alkyl,

R<sup>2</sup>, R<sup>3</sup>, and R<sup>4</sup>, are independently, C<sub>1-6</sub>alkyl, triC<sub>1-6</sub>alkylsiloxy, phenyl, naphthyl, substituted C<sub>1-6</sub>alkyl, substituted phenyl, or substituted naphthyl

where the alkyl substituents are selected from one or more members of the group consisting of C<sub>1-6</sub>alkoxycarbonyl, C<sub>1-6</sub>alkyl, C<sub>1-6</sub>alkoxy, amide, halogen, hydroxyl, carboxyl, C<sub>1-6</sub>alkylcarbonyl and formyl, and

5 where the aromatic substituents are selected from one or more members of the group consisting of C<sub>1-6</sub>alkoxycarbonyl, C<sub>1-6</sub>alkyl, C<sub>1-6</sub>alkoxy, amide, halogen, hydroxyl, carboxyl, C<sub>1-6</sub>alkylcarbonyl and formyl;

10 R<sup>5</sup> is a hydroxyl, an alkyl group containing one or more hydroxyl groups; or (CH<sub>2</sub>(CR<sup>9</sup>R<sup>10</sup>)<sub>y</sub>O)<sub>x</sub>-R<sup>11</sup> wherein y is 1 to 5, preferably 1 to 3, x is an integer of 1 to 100, preferably 2 to 90 and more preferably 10 to 25; R<sup>9</sup> - R<sup>11</sup> are independently selected from H, alkyl having up to 10 carbon atoms and alkyls having up to 10 carbon atoms substituted with at least one polar functional group,

15 R<sup>6</sup> is a divalent group comprising up to 20 carbon atoms;

R<sup>7</sup> is a monovalent group that can undergo free radical or cationic polymerization, comprising up to 20 carbon atoms; and

R<sup>8</sup> is a divalent or trivalent group comprising up to 20 carbon atoms.

20

7. The biomedical device of claim 1 wherein said hydroxyl-functionalized silicone-containing monomer is selected from the group consisting of 2-propenoic acid, 2-methyl-2-hydroxy-3-[3-[1,3,3,3-tetramethyl-1-(trimethylsilyl)oxy]disiloxanyl]propoxy]propyl ester, (3-methacryloxy-2-hydroxypropyloxy)propyltris(trimethylsiloxy)silane, (2-methacryloxy-3-hydroxypropyloxy)propylbis(trimethylsiloxy)methylsilane and mixtures thereof.

25

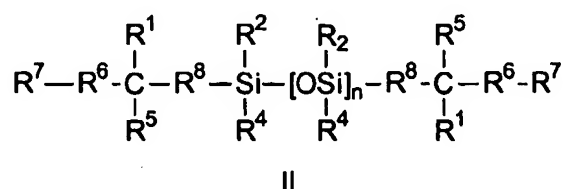
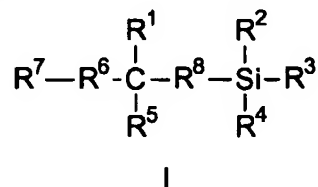
8. The biomedical device of claim 7 wherein the effective amount of said hydroxyl-functionalized silicone-containing monomer is about 20% to about 50%.

30

9. The biomedical device of claim 7 further comprising about 10% to about 40% additional silicone containing monomer, about 10% to about 50% hydrophilic monomers, and about 3% to about 15% high molecular weight hydrophilic polymer.

10. The biomedical device of claim 9, wherein the device is a soft contact lens.
11. The biomedical device of claim 1, wherein the hydroxyl-functionalized silicone-containing monomer has a monomer compatibility index of greater than about 0.5.
12. The biomedical device of claim 1, wherein the high molecular weight hydrophilic monomer is poly-N-vinylpyrrolidone.
13. A method of preparing a biomedical device comprising mixing a high molecular weight hydrophilic polymer and an effective amount of a hydroxyl-functionalized silicone-containing monomer to form a clear solution, and curing said solution.
14. A silicone hydrogel formed from a reaction mixture comprising a high molecular weight hydrophilic polymer and an effective amount of an hydroxyl-functionalized silicone-containing monomer.
15. 15. The silicone hydrogel of claim 14 wherein said hydrophilic polymer is selected from the group consisting of polyamides, polylactones, polyimides, polylactams, functionalized polyamides, functionalized polylactones, functionalized polyimides, functionalized polylactams, and mixtures thereof.
- 20 16. The silicone hydrogel of claim 14 wherein said hydrophilic polymer is selected from the group consisting of poly-N-vinyl pyrrolidone, poly-N-vinyl-2- piperidone, poly-N-vinyl-2-caprolactam, poly-N-vinyl-3-methyl-2- caprolactam, poly-N-vinyl-3-methyl-2-piperidone, poly-N-vinyl-4-methyl-2- piperidone, poly-N-vinyl-4-methyl-2-caprolactam, poly-N-vinyl-3-ethyl-2- pyrrolidone, and poly-N-vinyl-4,5-dimethyl-2-pyrrolidone, polyvinylimidazole, poly-N-N-dimethylacrylamide, polyvinyl alcohol, polyacrylic acid, polyethylene oxide, poly 2 ethyl oxazoline, heparin polysaccharides, polysaccharides, mixtures and copolymers thereof.
- 25 17. The silicone hydrogel of claim 14 wherein said hydrophilic polymer comprises poly-N-vinylpyrrolidone.
- 30 18. The silicone hydrogel of claim 14 comprising about 1 to about 15 weight% high molecular weight hydrophilic polymer

19. The silicone hydrogel of claim 14 comprising about 3 to about 15 weight% high molecular weight hydrophilic polymer.
20. The silicone hydrogel of claim 14 comprising about 5 to about 12 weight% high molecular weight hydrophilic polymer.
- 5 21. The silicone hydrogel of claim 14 wherein said hydroxyl-functionalized silicone-containing monomer is a compound of Formula I or II



wherein:

n is an integer between 3 and 35

15 R<sup>1</sup> is hydrogen, C<sub>1-6</sub>alkyl,

R<sup>2</sup>, R<sup>3</sup>, and R<sup>4</sup>, are independently, C<sub>1-6</sub>alkyl, triC<sub>1-6</sub>alkylsiloxy, phenyl, naphthyl, substituted C<sub>1-6</sub>alkyl, substituted phenyl, or substituted naphthyl

20 where the alkyl substituents are selected from one or more members of the group consisting of C<sub>1-6</sub>alkoxycarbonyl, C<sub>1-6</sub>alkyl, C<sub>1-6</sub>alkoxy, amide, halogen, hydroxyl, carboxyl, C<sub>1-6</sub>alkylcarbonyl and formyl, and

25 where the aromatic substituents are selected from one or more members of the group consisting of C<sub>1-6</sub>alkoxycarbonyl, C<sub>1-6</sub>alkyl, C<sub>1-6</sub>alkoxy, amide, halogen, hydroxyl, carboxyl, C<sub>1-6</sub>alkylcarbonyl and formyl;

R<sup>5</sup> is a hydroxyl, an alkyl group containing one or more hydroxyl groups, or (CH<sub>2</sub>(CR<sup>9</sup>R<sup>10</sup>)<sub>y</sub>O)<sub>x</sub>-R<sup>11</sup> wherein y is 1 to 5, preferably 1 to 3, x is an integer of 1 to 100, preferably 2 to 90 and more

preferably 10 to 25; R<sup>9</sup> - R<sup>11</sup> are independently selected from H, alkyl having up to 10 carbon atoms and alkyls having up to 10 carbon atoms substituted with at least one polar functional group,;

- 5        R<sup>6</sup> is a divalent group comprising up to 20 carbon atoms;  
       R<sup>7</sup> is a monovalent group that can undergo free radical or cationic polymerization, comprising up to 20 carbon atoms, and  
       R<sup>8</sup> is a divalent or trivalent group comprising up to 20 carbon atoms.
22.    The silicone hydrogel of claim 14 wherein the effective amount of  
 10       hydroxyl-functionalized silicone-containing monomer is about 20% to about 50%.
23.    The silicone hydrogel of claim 21 wherein said hydroxyl-functionalized  
       silicone-containing monomer is selected from the group consisting of  
 15       2-propenoic acid, 2-methyl-2-hydroxy-3-[3-[1,3,3,3-tetramethyl-1-  
       [trimethylsilyl]oxy]disiloxanyl]propoxy]propyl ester, (3-methacryloxy-2-  
       hydroxypropyloxy)propyltris(trimethylsiloxy)silane, (2-methacryloxy-3-  
       hydroxypropyloxy)propylbis(trimethylsiloxy)methylsilane and mixtures thereof.
24.    The silicone hydrogel of claim 14 further comprising at least one  
 20       additional silicone containing monomer.
25.    The silicone hydrogel of claim 24 wherein said additional silicone  
       containing monomers are selected from the group consisting of amide  
       analogs of 3-methacryloxypropyltris(trimethylsiloxy)silane; siloxane  
       vinyl carbamate analogs, siloxane vinyl carbonate analogs, and  
 25       siloxane containing monomers, combinations and oligomers thereof.
26.    The silicone hydrogel of claim 24 wherein said additional silicone  
       containing monomers are selected from the group consisting of  
       3-methacryloxypropyltris(trimethylsiloxy)silane,  
       monomethacryloxypropyl terminated polydimethylsiloxanes,  
 30       polydimethylsiloxanes,  
       3-methacryloxypropylbis(trimethylsiloxy)methylsilane,  
       methacryloxypropylpentamethyl disiloxane and combinations thereof.

27. The silicone hydrogel of claim 24 wherein said additional silicone containing monomers are present in an amount of 0 to about 80 weight %.
28. The silicone hydrogel of claim 24 wherein said additional silicone containing monomers are present in an amount of about 5 to about 60%.
29. The silicone hydrogel of claim 24 wherein said additional silicone containing monomers are present in an amount of about 10 to about 40%.
30. The silicone hydrogel of claim 14 further comprising at least one hydrophilic monomer.
31. The silicone hydrogel of claim 30 wherein said at least one hydrophilic monomer comprises at least one acrylic group, vinyl group or a combination thereof.
32. The silicone hydrogel of claim 31 wherein said acrylic group has the formula  $\text{CH}_2=\text{CRCOX}$ , where R is hydrogen or  $\text{C}_{1-6}$ alkyl and X is O or N.
33. The silicone hydrogel of claim 30 wherein said at least one hydrophilic monomer is selected from the group consisting of N,N-dimethylacrylamide, 2-hydroxyethyl methacrylate, glycerol methacrylate, 2-hydroxyethyl methacrylamide, polyethyleneglycol monomethacrylate, methacrylic acid, acrylic acid, N-vinyl pyrrolidone, N-vinyl-N-methyl acetamide, N-vinyl-N-ethyl acetamide, N-vinyl-N-ethyl formamide, N-vinyl formamide, hydrophilic vinyl carbonate monomers, vinyl carbamate monomers, hydrophilic oxazolone monomers, polydextran and copolymers and combinations thereof.
34. The silicone hydrogel of claim 30 wherein said at least one hydrophilic monomer comprises at least one polyoxyethylene polyol having one or more of the terminal hydroxyl groups replaced with a functional group containing a polymerizable double bond.
35. The silicone hydrogel of claim 34 wherein said at least one hydrophilic monomer is selected from the group consisting of polyethylene glycol, ethoxylated alkyl glucoside, and polyethylene polyols having one or

more terminal polymerizable olefinic groups bonded to the polyethylene polyol.

36. The silicone hydrogel of claim 30 wherein said at least one hydrophilic monomer is selected from the group consisting of N,N-  
5 dimethylacrylamide, 2-hydroxyethyl methacrylate, glycerol methacrylate, 2-hydroxyethyl methacrylamide, N-vinylpyrrolidone, polyethyleneglycol monomethacrylate, methacrylic acid, acrylic acid and combinations thereof.
37. The silicone hydrogel of claim 30 wherein said at least one hydrophilic  
10 monomer comprises N,N-dimethylacrylamide.
38. The silicone hydrogel of claim 30 wherein said at least one hydrophilic monomer is present in amounts of about 0 to about 70 weight%.
39. The silicone hydrogel of claim 30 wherein said at least one hydrophilic monomer is present in amounts of about 5 to about 60 weight%.
- 15 40. The silicone hydrogel of claim 30 wherein said at least one hydrophilic monomer is present in amounts of about 10 to 50 weight%.
41. The silicone hydrogel of claim 14 comprising about 1 to about 15 weight% high molecular weight hydrophilic polymer and about 5 to about 90 weight % hydroxyl-functionalized silicone-containing  
20 monomer.
42. The silicone hydrogel of claim 14 comprising about 3 to about 15 weight% high molecular weight hydrophilic polymer and about 10 to about 80 weight % hydroxyl-functionalized silicone-containing monomer.
- 25 43. The silicone hydrogel of claim 15 comprising about 5 to about 12 weight% high molecular weight hydrophilic polymer and about 20 to about 50 weight % hydroxyl-functionalized silicone-containing monomer.
44. The silicone hydrogel of claim 14 further comprising 0 to about 80  
30 weight % at least one additional silicone containing monomer.
45. The silicone hydrogel of claim 14 further comprising about 0 to about 70 weight % at least one hydrophilic monomer.
46. The silicone hydrogel of claim 44 comprising about 5 to about 60 weight % of said at least one additional silicone containing monomer.

47. The silicone hydrogel of claim 44 comprising about 10 to about 40 weight % of said at least one additional silicone containing monomer.
48. The silicone hydrogel of claim 45 comprising about 5 to about 60 weight % of said at least one hydrophilic monomer.
- 5 49. The silicone hydrogel of claim 45 comprising about 10 to about 50 weight % of said at least one hydrophilic monomer.
50. The silicone hydrogel of claim 14 comprising about 1% to about 15% high molecular weight hydrophilic polymer; about 5 to about 90 weight% hydroxyl-functionalized silicone containing monomer; 5 to  
10 about 60 weight% additional silicone containing monomer and 5 to about 60 weight % hydrophilic monomer.
51. The silicone hydrogel of claim 14 comprising about 3% to about 15% high molecular weight hydrophilic polymer; about 10 to about 80 weight% hydroxyl-functionalized silicone containing monomer; 5 to  
15 about 60 weight% additional silicone containing monomer and 5 to about 60 weight % hydrophilic monomer.
52. The silicone hydrogel of claim 14 comprising about 5% to about 12% high molecular weight hydrophilic polymer; about 20 to about 50 weight% hydroxyl-functionalized silicone containing monomer; 5 to  
20 about 60 weight% additional silicone containing monomer and 50 to about 60 weight % hydrophilic monomer.
53. The silicone hydrogel of claim 14 comprising about 5% to about 12% high molecular weight hydrophilic polymer; about 20 to about 50 weight% hydroxyl-functionalized silicone containing monomer; 10 to  
25 about 40 weight% additional silicone containing monomer and 10 to about 50 weight % hydrophilic monomer.
54. The silicone hydrogel of claim 14 wherein said hydrophilic polymer has a weight average molecular weight of at least about 10,000 daltons.
55. The silicone hydrogel of claim 14 wherein said hydrophilic polymer  
30 comprises at least one cyclic moiety along a backbone portion of said hydrophilic polymer.
56. The silicone hydrogel of claim 14 wherein said hydrophilic polymer is selected from the group consisting of polymers comprising cyclic amides, cyclic imides and combinations thereof.

57. An ophthalmic device comprising the silicone hydrogel of claim 14.
58. The ophthalmic I device of claim 57 further comprising an advancing dynamic contact angle of less than about 70°.
59. The ophthalmic I device of claim 57 further comprising an advancing  
5 dynamic contact angle of less than about 60°.
60. The ophthalmic device of claim 57 further comprising, after about one day of wear, a tear film break up time of at least about 7 seconds and an oxygen permeability of at least about 50 barrers.
61. The ophthalmic device of claim 58 further comprising a modulus of  
10 less than about 90 psi.
62. The ophthalmic device of claim 58 further comprising a water content between about 10 and about 60%.
63. The ophthalmic device of claim 58 further comprising a water content between about 20 and about 55%.
- 15 64. The ophthalmic device of claims 58 further comprising a water content between about 25 and about 40%.
65. The ophthalmic device of claim 60 wherein said tear film breakup time is at least about 7 seconds after about 7 days of wear.
66. A method comprising the steps of (a) mixing reactive components  
20 comprising at least one high molecular weight hydrophilic polymer and an effective amount of at least one hydroxyl-functionalized silicone-containing monomer and (b) curing the product of step (a) to form a biomedical device.
67. The method of claim 66 wherein said biomedical device comprises an  
25 ophthalmic device.
68. The method of claim 67 wherein said ophthalmic device is a silicone hydrogel contact lens.
69. The method of claim 67 wherein said reactive components are mixed in the presence of a diluent to form a reaction mixture.
- 30 70. The method of claim 69 wherein said diluent is selected from the group consisting of ethers, esters, alkanes, alkyl halides, silanes, amides, alcohols and mixtures thereof.

71. The method of claim 69 wherein said diluent selected from the group consisting amides, alcohols and mixtures thereof.
72. The method of claim 69 wherein said diluent selecting the group consisting of secondary and tertiary alcohols and mixtures thereof.
- 5 73. The method of claim 69 wherein said diluent selecting the group consisting of tetrahydrofuran, ethyl acetate, methyl lactate, i-propyl lactate, ethylene chloride, octamethylcyclotetrasiloxane, dimethyl formamide, dimethyl acetamide, dimethyl propionamide, N methyl pyrrolidinone mixtures thereof and mixtures of any of the foregoing with  
10 at least one alcohol.
74. The method of claim 71 wherein said diluent comprises at least one alcohol having at least 5 carbon atoms.
75. The method of claim 71 wherein said diluent comprises at least one alcohol having at least 6 carbons atoms.
- 15 76. The method of claim 69 wherein said diluents are inert and easily displaceable with water.
77. The method of claim 69 wherein said diluent comprises at least one alcohol selected from the group consisting of tert-butanol, tert-amyl alcohol, 2-butanol, 2-methyl-2-pentanol, 2,3-dimethyl-2-butanol, 3-  
20 methyl-3-pentanol, 3-ethyl-3-pentanol, 3,7-dimethyl-3-octanol and mixtures thereof.
78. The method of claim 69 wherein said diluent is selected from the group consisting of hexanol, heptanol, octanol, nonanol, decanol, tert-butyl alcohol, 3-methyl-3-pentanol, isopropanol, t amyl alcohol, ethyl lactate,  
25 methyl lactate, i-propyl lactate, 3,7-dimethyl-3-octanol, dimethyl formamide, dimethyl acetamide, dimethyl propionamide, N methyl pyrrolidinone and mixtures thereof.
79. The method of claim 69 wherein said diluent is selected from the group consisting of 1-ethoxy-2-propanol, 1-methyl-2-propanol, t-amyl alcohol, tripropylene glycol methyl ether, isopropanol, 1-methyl-2-pyrrolidone,  
30 ethyl lactate, N,N-dimethylpropionamide, dipropylene glycol methyl ether and mixtures thereof.
80. The method of claim 69 wherein said diluent is present in an amount less than about 50 weight %, based upon the reaction mixture.

81. The method of claim 69 wherein said diluent is present in an amount less than about 40 weight% based upon the reaction mixture.
82. The method of claim 69 wherein said diluent is present in an amount between about 10 and about 30 weight% based upon the reaction mixture.
83. The method of claim 69 wherein said diluent is water soluble at processing conditions and said process further comprises (c) removing said lens from a mold in which said lens was cured and (d) hydrating said lens, wherein both steps (c) and (d) are performed in aqueous solutions which comprise water as a substantial component.
84. The method of claim 69 wherein said reaction mixture further comprises at least one initiator.
85. The method of claim 84 wherein said curing is conducted via heat, exposure to radiation or a combination thereof.
86. The method of claim 84 wherein said curing is conducted via irradiation comprises ionizing and/or actinic radiation and said initiator comprises at least one photoinitiator.
87. The method of claim 86 wherein said radiation comprises light having a wavelength of about 200 to about 500nm and said initiator is selected from the group consisting of aromatic alpha-hydroxy ketones, alkoxyoxybenzoin, acetophenones, acyl phosphine oxides, mixtures of tertiary amines and diketones, and mixtures thereof.
88. The method of claim 86 wherein said initiator is selected from the group consisting of 1-hydroxycyclohexyl phenyl ketone, 2-hydroxy-2-methyl-1-phenyl-propan-1-one, bis(2,6-dimethoxybenzoyl)-2,4,4-trimethylpentyl phosphine oxide, bis(2,4,6-trimethylbenzoyl)-phenylphosphineoxide, 2,4,6-trimethylbenzoyldiphenyl phosphine oxide and 2,4,6-trimethylbenzoyl diphenylphosphine oxide, benzoin methyl ester, combinations of camphorquinone and ethyl 4-(N,N-dimethylamino)benzoate and mixtures thereof.
89. The method of claim 84 wherein said initiator is present in the reaction mixture in amounts effective to initiate polymerization of the reaction mixture.

90. The method of claim 84 wherein said initiator is present in the reaction mixture in amounts from about 0.1 to about 2 weight percent based upon said reactive components.
- 5 91. The method of claim 86 wherein said curing is conducted via visible light irradiation.
92. The method of claim 93 wherein said initiator comprises 1-hydroxycyclohexyl phenyl ketone, bis(2,6-dimethoxybenzoyl)-2,4,4-trimethylpentyl phosphine oxide and mixtures thereof.
- 10 93. The method of claim 93 wherein said initiator comprises bis(2,4,6-trimethylbenzoyl)-phenylphosphineoxide.
94. The method of claims 93 wherein said reactive components further comprises at least one UV absorbing compound.
95. The method of claim 91 wherein said curing step is conducted at a cure intensity between about 0.1 and about 6 mW/cm<sup>2</sup>.
- 15 96. The method of claim 91 wherein said curing step is conducted at a cure intensity of between about 0.2 mW/cm<sup>2</sup> to about 3 mW/cm<sup>2</sup>.
97. The method of claims 95 wherein said curing step further comprises a cure time of at least about 1 minute.
98. The method of claims 95 wherein said curing step further comprises a  
20 cure time of between about 1 and about 60 minutes.
99. The method of claim 95 wherein said curing step further comprises a cure time of between about 1 and about 30 minutes.
100. The method of claim 97 wherein said curing step is conduct at a temperature greater than about 25°C.
- 25 101. The method of claim 97 wherein said curing step is conduct at a temperature between about 25°C and 70°C.
102. The method of claim 97 wherein said curing step is conduct at a temperature between about 40°C and 70°C.
103. The method of claim 69 wherein said reaction mixture is cured in a  
30 mold and said method further comprises the step deblocking said ophthalmic device from said mold.
104. The method of claim 103 wherein said reaction mixture further comprises at least one low molecular weight hydrophilic polymer.

105. The method of claim 104 wherein said low molecular weight hydrophilic polymer has a number average molecular weight of less than about 40,000 Dalton.
106. The method of claim 104 wherein said low molecular weight hydrophilic polymer has a number average molecular weight of less than about 20,000 Dalton.
107. The method of claim 104 wherein the low molecular weight polymer is selected from the group consisting of water soluble polyamides, lactams and polyethylene glycols, and mixtures thereof.
108. The method of claim 104 wherein the low molecular weight polymer is selected from the group consisting poly-vinylpyrrolidone, polyethylene glycols, poly 2 ethyl-2-oxazoline and mixtures thereof.
109. The method of claim 104 wherein the low molecular weight hydrophilic polymer is present in amounts up to about 20 weight% based upon the reaction mixture.
110. The method of claim 104 wherein the low molecular weight hydrophilic polymer is present in amounts between about 5 and about 20 weight% based upon the reaction mixture.
111. The method of claim 103 wherein said deblocking is conducted using an aqueous solution.
112. The method of claim 103 wherein said aqueous solution further comprises at least one surfactant.
113. The method of claim 112 wherein said surfactant comprises at least one non-ionic surfactant.
114. The method of claim 112 wherein said surfactant comprises TWEEN®, or DOE120.
115. The method of claim 112 wherein said surfactant is present in amounts up to about 10,000 ppm.
116. The method of claim 112 wherein said surfactant is present in amounts between about 500 ppm and about 1500 ppm.
117. The method of claim 112 wherein said surfactant is present in amounts between about 100 and about 1200 ppm.
118. The method of claim 103 wherein said aqueous solution comprises at least one organic solvent.

119. The method of claim 103 wherein said deblocking is conducted at a temperature between about ambient and about 100°C.
120. The method of claim 103 wherein said deblocking is conducted at a temperature between about 70°C and about 95°C.
- 5 121. The method of claim 103 wherein said deblocking is conducted using agitation.
122. The method of claim 103 wherein said agitation comprises sonication.
123. A method comprising the steps of (a) mixing reactive components comprising a high molecular weight hydrophilic polymer and an effective amount of a hydroxyl-functionalized silicone containing monomer and (b) curing the product of step (a) at or above a minimum gel time, to form a wettable biomedical device.
- 10 124. The method of claim 123 wherein said device is a ophthalmic lens.
125. The method of claim 123 wherein said device is a contact lens.
- 15 126. The method of claim 125 wherein said wettability comprises an advancing dynamic contact angle of about 70° or less.
127. The method of claim 125 wherein said wettability comprises an advancing dynamic contact angle of about 60° or less.
128. The method of claim 125 wherein said wettability comprises a tear film break up time of at least about 7 seconds.
- 20 129. The method of claim 125 wherein said reactive components further comprises at least one initiator
130. The method of claim 129 wherein said cure is conducted via irradiation and said conditions comprise an initiator concentration and cure intensity effective to provide said minimum gel time.
- 25 131. The method of claim 129 wherein said initiator is present in an amount up to about 1% based upon all reactive components.
132. The method of claim 129 wherein said initiator is present in an amount less than about 0.5% based upon all reactive components.
- 30 133. The method of claim 130 wherein said cure is conducted via irradiation at an intensity of less than about 5 mW/cm<sup>2</sup>.
134. The method of claim 130 wherein said gel time is at least about 30 seconds.

135. The method of claim 123 wherein said gel time is at least about 35 seconds.
136. An ophthalmic device comprising at least one hydroxyl-functionalized silicone-containing monomer and an amount of high molecular weight hydrophilic polymer sufficient to provide said device, without a surface treatment, with an advancing dynamic contact angle of less than about 80°.
137. An ophthalmic device comprising at least one hydroxyl-functionalized silicone-containing monomer and an amount of high molecular weight hydrophilic polymer sufficient to provide said device, without a surface treatment, with an advancing dynamic contact angle which is at least about 10% lower than an advancing dynamic contact angle for an ophthalmic device which does not contain said hydrophilic polymer.
138. An ophthalmic device comprising at least one hydroxyl-functionalized silicone-containing monomer and an amount of high molecular weight hydrophilic polymer sufficient to provide said device, without a surface treatment, with tear film break up time after about one day of wear of at least about 7 seconds.
139. The ophthalmic device claim 136 wherein said device is a silicone hydrogel contact lens.
140. The ophthalmic device of claim 136 wherein said hydroxyl-functionalized silicone containing monomer is present in a compatibilizing effective amount.
141. The ophthalmic device of claim 136 wherein said hydroxyl-functionalized silicone containing monomer is present in an amount between about 5% to about 90%
142. The biomedical device of claim 136 comprising about 1% to about 15% high molecular weight hydrophilic polymer.
143. The biomedical device of claim 136 further comprising 0 to about 80 weight % at least one additional silicone containing monomer.
144. The biomedical device of claim 136 further comprising 0 to about 70 weight % at least one hydrophilic monomer.

145. The biomedical device of claim 136 comprising about 1% to about 15% high molecular weight hydrophilic polymer; about 5 to about 90 weight% hydroxyl-functionalized silicone containing monomer; 0 to about 80 weight% additional silicone containing monomer and 0 to about 70 weight % hydrophilic monomer.
146. The biomedical device of claim 136 further comprising a haze value of less than about 100.
147. The biomedical device of claim 136 further comprising a haze value of less than about 50.
148. The silicone hydrogel of claim 50 wherein said hydroxyl-functionalized silicone containing monomer comprises 2-propenoic acid, 2-methyl-,2-hydroxy-3-[3-[1,3,3,3-tetramethyl-1-[(trimethylsilyl)oxy]disiloxanyl]propoxy]propyl ester, said hydrophilic polymer comprises polyvinylpyrrolidone; said additional silicone containing monomer comprises monomethacryloxypropyl terminated mono-n-butyl terminated polydimethylsiloxane and said hydrophilic monomers comprise 2-hydroxyethyl methacrylate and N,N-dimethylacrylamide.
149. The silicone hydrogel of claim 148 comprising the reaction product of about 28 weight % 2-propenoic acid, 2-methyl-,2-hydroxy-3-[3-[1,3,3,3-tetramethyl-1-[(trimethylsilyl)oxy]disiloxanyl]propoxy]propyl ester, about 31 weight % monomethacryloxypropyl terminated mono-n-butyl terminated polydimethylsiloxane about 24 weight % N,N-dimethylacrylamide, about 6 weight % 2-hydroxyethyl methacrylate, about 1.5 weight % tetraethyleneglycoldimethacrylate, about 7 weight % polyvinylpyrrolidone, and up to about 1 weight % photoinitiator.
150. The silicone hydrogel of claim 148 comprising the reaction product of about 30 weight % 2-propenoic acid, 2-methyl-,2-hydroxy-3-[3-[1,3,3,3-tetramethyl-1-[(trimethylsilyl)oxy]disiloxanyl]propoxy]propyl ester, about 23 weight % monomethacryloxypropyl terminated mono-n-butyl terminated polydimethylsiloxane about 31 weight % N,N-dimethylacrylamide, about 7.5 weight % 2-hydroxyethyl methacrylate, about 0.75 weight % ethyleneglycoldimethacrylate, about 6 weight % polyvinylpyrrolidone, and up to about 1 weight % photoinitiator.

151. The silicone hydrogel of claim 148 comprising the reaction product of about 30 weight % 2-propenoic acid, 2-methyl-,2-hydroxy-3-[3-[1,3,3,3-tetramethyl-1-[(trimethylsilyl)oxy]disiloxanyl]propoxy]propyl ester, about 18 weight % monomethacryloxypropyl terminated mono-n-butyl terminated polydimethylsiloxane about 31 weight % N,N-dimethylacrylamide, about 9 weight % 2-hydroxyethyl methacrylate, about 0.8 weight % ethyleneglycoldimethacrylate, about 6 weight % polyvinylpyrrolidone, and less than about 1 weight % photoinitiator.
152. A method for improving the wettability of an ophthalmic device formed from a reaction mixture comprising adding at least one high molecular weight hydrophilic polymer and an effective amount of at least one compatibilizing monomer to said reaction mixture.
153. The method of claim 152 wherein said compatibilizing monomer has a compatibility index of greater than about 0.5.
154. The method of claim 152 wherein said compatibilizing monomer has a compatibility index of greater than about 1.
155. The method of claim 152 wherein said compatibilizing monomer comprises at least one siloxane group.
156. A method for improving the wettability of an ophthalmic device formed from a reaction mixture comprising incorporating at least one high molecular weight hydrophilic polymer and an effective amount of at least one hydroxyl-functionalized silicone containing monomer to said reaction mixture.
157. The method of claim 156 wherein said compatibilizing monomer has a compatibility index of greater than about 0.5.
158. The method of claim 156 wherein said compatibilizing monomer has a compatibility index of greater than about 1.
159. The method of claim 156 wherein said compatibilizing monomer has a Si to OH ratio of less than about 15:1.
160. The method of claim 156 wherein said compatibilizing monomer has a Si to OH ratio of between about 1:1 to about 10:1
161. The method of claim 125 wherein said wettability comprises a contact angle of about 80° or less.

162. A device comprising a silicone hydrogel contact lens which is substantially free from surface deposition without surface modification.

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**WO 03/022321 A3**

(54) Title: **BIOMEDICAL DEVICES CONTAINING INTERNAL WETTING AGENTS**

(57) Abstract: This invention includes a wettable biomedical device containing a high molecular weight hydrophilic polymer and a hydroxyl-functionalized silicone-containing monomer.

## INTERNATIONAL SEARCH REPORT

Internatio plication No

PCT/US 02/28613

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 A61L27/52 A61L27/26 A61L27/36 A61L31/04 A61F2/14  
G02C7/04

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 A61L A61F G02C

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

WPI Data, EPO-Internal, COMPENDEX, EMBASE, CHEM ABS Data, BIOSIS, PAJ

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 4 136 250 A (MUELLER KARL F ET AL) 23 January 1979 (1979-01-23) claims ---	1-151, 156-161
X	KUNZLER J F: "Silicone Hydrogels for Contact Lens Application" TRENDS IN POLYMER SCIENCE, ELSEVIER SCIENCE PUBLISHERS B.V. AMSTERDAM, NL, vol. 4, no. 2, 1 February 1996 (1996-02-01), pages 52-59, XP004049314 ISSN: 0966-4793 page 53, right-hand column, line 55 -page 54, left-hand column, line 9 figure 1 --- -/--	1-151, 156-161



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

## \* Special categories of cited documents:

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\*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

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## INTERNATIONAL SEARCH REPORT

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PCT/US 02/28613

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 6 218 503 B1 (LAI YU-CHIN ET AL) 17 April 2001 (2001-04-17) column 6, line 18 -column 7, line 9 examples 3,6,14 claims ---	1-151, 156-161
X	US 4 139 692 A (TANAKA KYOICHI ET AL) 13 February 1979 (1979-02-13) column 5, line 22 - line 56 claims ---	1-151, 156-161
X	EP 0 940 693 A (JOHNSON & JOHNSON VISION PROD) 8 September 1999 (1999-09-08) page 3, line 30 -page 4, line 10 claims ---	1-151, 156-161
X	WO 01 27174 A (JOHNSON & JOHNSON VISION CARE) 19 April 2001 (2001-04-19) page 3, line 20 -page 4, line 25 page 5, line 14 -page 6, line 5 claims ---	1-151, 156-161
P,X	WO 01 70837 A (JOHNSON & JOHNSON VISION CARE) 27 September 2001 (2001-09-27) page 3, line 3 -page 4, line 5 page 8, line 5 - line 31 claims ---	1-151, 156-161
X	EP 0 989 418 A (JOHNSON & JOHNSON VISION PROD) 29 March 2000 (2000-03-29) page 2, line 42 - line 48 page 3, line 13 - line 18 page 3, line 41 -page 4, line 12 table 2 claims ---	152-155
X	US 6 087 415 A (TURNER DAVID C ET AL) 11 July 2000 (2000-07-11) abstract examples 1,5 table 1 ---	152-155
X	US 4 487 905 A (MITCHELL DARRELL D) 11 December 1984 (1984-12-11) abstract column 4, line 10 - line 20 column 12, line 29 -column 13, line 10 examples 1,2 table 1 claims ---	152-155
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## INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 02/28613

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 92 18548 A (OPTICAL RESEARCH INC) 29 October 1992 (1992-10-29) page 1, line 1 - line 8 page 16, line 19 - line 27 claim 1 -----	162
X	WO 97 20852 A (BAUSCH & LOMB) 12 June 1997 (1997-06-12) page 2, line 10 - line 27 page 3, line 25 - line 30 claims 1,18 -----	162

# INTERNATIONAL SEARCH REPORT

Intern application No.  
PCT/US 02/28613

## Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This International Search Report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:  
because they relate to subject matter not required to be searched by this Authority, namely:
  
2. ☐ Claims Nos.:  
because they relate to parts of the International Application that do not comply with the prescribed requirements to such an extent that no meaningful International Search can be carried out, specifically:
  
3. ☐ Claims Nos.:  
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

## Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

see additional sheet

1. ☒ As all required additional search fees were timely paid by the applicant, this International Search Report covers all searchable claims.
  
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
  
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this International Search Report covers only those claims for which fees were paid, specifically claims Nos.:
  
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this International Search Report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.
- ☒ No protest accompanied the payment of additional search fees.

**FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210**

This International Searching Authority found multiple (groups of) inventions in this international application, as follows:

**1. Claims: 1-151, 156-161**

A silicone hydrogel formed from a reaction mixture comprising a HIGH MOLECULAR WEIGHT HYDROPHILIC POLYMER and an effective amount of an HYDROXYL-FUNCTIONALIZED SILICONE-CONTAINING MONOMER. Methods to produce such an hydrogel as well as specific products made of such hydrogels and the methods to produce them.

**2. Claims: 152-155**

A method for improving the wettability of an ophthalmic device formed from a reaction mixture comprising adding at least one HIGH MOLECULAR WEIGHT HYDROPHILIC POLYMER and an effective amount of at least one COMPATIBILIZING MONOMER to said reaction mixture.

**3. Claim : 162**

A device comprising a silicone hydrogel contact lens which is substantially free from surface deposition without surface modification.

## INTERNATIONAL SEARCH REPORT

Information on patent family members

International Publication No

PCT/US 02/28613

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
US 4136250	A	23-01-1979	DE 2831274 A1	08-02-1979
			FR 2398084 A1	16-02-1979
			GB 2002788 A , B	28-02-1979
			JP 1430546 C	24-03-1988
			JP 54022487 A	20-02-1979
			JP 62036046 B	05-08-1987
US 6218503	B1	17-04-2001	US 6268467 B1	31-07-2001
			US 6008317 A	28-12-1999
			AU 3645699 A	06-12-1999
			DE 69904468 D1	23-01-2003
			EP 1080117 A1	07-03-2001
			JP 2002515519 T	28-05-2002
			WO 9960029 A1	25-11-1999
US 4139692	A	13-02-1979	JP 1093651 C	27-04-1982
			JP 54055455 A	02-05-1979
			JP 56039450 B	12-09-1981
			AU 528355 B2	28-04-1983
			AU 4036078 A	17-04-1980
			CA 1103838 A1	23-06-1981
			DE 2844078 A1	03-05-1979
			FR 2405960 A1	11-05-1979
			GB 2009189 A , B	13-06-1979
EP 0940693	A	08-09-1999	US 5998498 A	07-12-1999
			AU 1848999 A	16-09-1999
			BR 9900828 A	23-05-2000
			CN 1234522 A	10-11-1999
			EP 0940693 A2	08-09-1999
			JP 11316358 A	16-11-1999
			KR 2000034800 A	26-06-2000
			SG 73617 A1	20-06-2000
			US 6367929 B1	09-04-2002
			US 2002107324 A1	08-08-2002
WO 0127174	A	19-04-2001	AU 7367600 A	23-04-2001
			BR 0014552 A	20-08-2002
			EP 1235866 A1	04-09-2002
			WO 0127174 A1	19-04-2001
			US 6367929 B1	09-04-2002
			US 2002107324 A1	08-08-2002
WO 0170837	A	27-09-2001	US 6367929 B1	09-04-2002
			AU 4765601 A	03-10-2001
			EP 1268590 A1	02-01-2003
			WO 0170837 A1	27-09-2001
EP 0989418	A	29-03-2000	US 6099852 A	08-08-2000
			AU 4876799 A	30-03-2000
			BR 9904255 A	17-04-2001
			CN 1249443 A	05-04-2000
			EP 0989418 A2	29-03-2000
			JP 2000137195 A	16-05-2000
			KR 2000034933 A	26-06-2000
			SG 85676 A1	15-01-2002
US 6087415	A	11-07-2000	AU 3223699 A	23-12-1999

## INTERNATIONAL SEARCH REPORT

Information on patent family members

International Publication No

PCT/US 02/28613

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 6087415	A	BR 9902243 A CN 1243846 A EP 0963761 A1 JP 2000070357 A KR 2000006070 A SG 87032 A1 US 6500481 B1	26-09-2000 09-02-2000 15-12-1999 07-03-2000 25-01-2000 19-03-2002 31-12-2002
US 4487905	A 11-12-1984	CA 1215797 A1 DE 3475629 D1 EP 0122061 A2 JP 1022306 B JP 1537995 C JP 59176350 A	23-12-1986 19-01-1989 17-10-1984 26-04-1989 16-01-1990 05-10-1984
WO 9218548	A 29-10-1992	CA 2108682 A1 WO 9218548 A1	19-10-1992 29-10-1992
WO 9720852	A 12-06-1997	AU 713509 B2 AU 1076597 A BR 9612119 A CA 2239901 A1 CN 1222155 A ,B DE 69615393 D1 DE 69615393 T2 EP 0865444 A1 ES 2164265 T3 JP 2001518061 T WO 9720852 A1 US 5710302 A US 5908906 A	02-12-1999 27-06-1997 17-02-1999 12-06-1997 07-07-1999 25-10-2001 04-07-2002 23-09-1998 16-02-2002 09-10-2001 12-06-1997 20-01-1998 01-06-1999